



Investigation of photodetector and photovoltaic properties of H₂Pc/CuO and H₂Pc/ITO junctions produced with modified H₂Pc

Fatih Unal^{1,a)} , Sitki Aktas^{2,a)}, Mustafa Sukru Kurt³, Mümin Mehmet Koc^{4,5}, Tayfun Arslan⁶, Burhan Coskun⁵, Mahmut Gur⁷

¹ Central Research Laboratories, Application and Research Centre, GRUMLAB, Giresun University, 28200 Giresun, Turkey

² Department of Mechanical Engineering, Faculty of Engineering, Giresun University, Giresun, Turkey

³ Department of Fundamental Science, Science Faculty, Erzurum Technical University, Erzurum, Turkey

⁴ Department of Medical Service and Techniques, School of Medical Service, Kırklareli University, Kırklareli, Turkey

⁵ Department of Physics, Faculty of Science and Literature, Kırklareli University, Kırklareli, Turkey

⁶ Department of Chemistry, Faculty of Science, Giresun University, Giresun, Turkey

⁷ Department of Forest Industrial Engineering, Faculty of Forestry, Kastamonu University, 37100 Kastamonu, Turkey

^{a)} Address all correspondence to these authors. e-mails: fatih.unal@giresun.edu.tr; sitki.aktas@giresun.edu.tr

Received: 20 April 2023; accepted: 29 November 2023; published online: 18 December 2023

In this work, it was aimed to provide an alternative to organic-based photodiode detectors (OPDs), solar cell with the modified H₂Pc organic thin film heterojunctions. For this purpose, the modified H₂Pc organic thin film was grown on ITO/glass and CuO_x/ITO/glass substrates via thermal evaporation method and their optoelectronic properties were compared with each other and other organic-based devices. The crucial optical parameters of the fabricated heterojunctions were determined in the wavelengths range of 300–800 nm. In order to perform electrical characterisation, Ag metal contacts were grown on the heterojunctions and Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices were fabricated. Diode, photovoltaic and photodetector parameters of the fabricated devices were investigated in the applied potential of ± 3 V under dark and several light intensities. Ultimately, it can be stated that the fabricated devices may offer an alternative to OPDs and organic-based solar cell.

Introduction

The use of organic semiconductors in a wide range of new technologies has attracted intense interest due to their structural and functional tunability. The combination of the film-forming ability of organics and the superior electrical properties of inorganic materials makes it possible to develop a new class of semiconductor devices which can be used as flexible electronics and optoelectronics and photovoltaics [1, 2]. Among these, using organic materials as an interface layer at metal–semiconductor (MS) structure would ensure beneficial improvements of the electrical properties of the structure. For instances, symmetrical CuPc organic molecule was coated on a p-Si wafer as an interfacial layer and it was obtained that Al/symmetrical CuPc/p-Si/Al heterojunction could be used as a photodevice or photosensor [3]. The negative differential resistance (NDR) behaviour was investigated on PMITz/ITO junction and it was seen that NDR effect strongly related with light intensity [4].

P3HT/n-Si organic–inorganic hybrid solar cell was fabricated and light dependence of the current transport mechanisms was investigated [2]. Coronene (C₂₄H₁₂) thin films were grown on the CdO [5], ZnO [6], Cu-doped ZnO [7] metal oxides structure and optoelectronic properties were analysed. A self-powered PEDOT:PSS/Ga₂O₃ organic–inorganic hybrid heterojunction was investigated with excellent photoelectric properties [8]. Additionally, Phthalocyanine (Pcs) molecules with a two-dimensional conjugated π -electrons system provide a large charge transfer [9]. Thanks to these properties, Pcs and their analogues have been widely used in several technological areas [10, 11]. Especially, the metal-free phthalocyanine (H₂Pc) materials are widely used in photodetectors, solar cells and other optoelectronic devices [12, 13].

Thin film copper oxides produced by several techniques are widely used metal oxide semiconductor in the area of device manufacturing, optoelectronic and photovoltaic

applications [14–16]. The oxide phases of copper oxide, which have distinct the optical and electrical properties, can be altered by reactive sputtering during the magnetron sputtering deposition process or by post-deposition annealing at various temperatures under vacuum or atmosphere [14, 17–20]. Hence, in this study, Cu thin film was grown on ITO substrate via magnetron sputtering technique and annealed in oxygen environment under vacuum conditions after deposition. Structural, morphological properties of deposited copper oxide thin film were investigated. Afterwards, unlike the known phthalocyanines, in order to investigate the interface layer performance of metal free-Pcs have chalcone, which π -bridge consists of α , β unsaturated carbonyl unit which is responsible for intramolecular charge transfer, the H_2Pc organic molecules was deposited as a thin film on commercial ITO and magnetron sputtered CuO_x surfaces via physical vapour deposition (PVD) method. Optical characterisation of H_2Pc /ITO and H_2Pc/CuO_x /ITO heterojunctions was examined. Eventually, with growing Ag contacts on both the heterojunctions, Ag/ H_2Pc /ITO/Ag and Ag/ H_2Pc / CuO_x /Ag devices were fabricated and photodiode, photodetector and photovoltaic characteristics of these devices were examined in detail.

Results and discussion

Structural, morphological and optical characterisation of CuO_x thin film

The XRD pattern of CuO_x thin film is presented in Fig. 1(a). It can be clearly observed that post-annealing at 200 °C during 30 min led to the copper thin film crystallised in the form of CuO_2 and cupric oxide (CuO). The most intense peak at 2θ of 35.22° represents the crystallographic reflection of (002) plane of monoclinic CuO with the space group $C12/c1(15)$ and the lattice constant of $a = 4.6839 \text{ \AA}$, $b = 3.4734 \text{ \AA}$, $c = 5.1226 \text{ \AA}$ and $\beta = 98.730^\circ$ (entry # 96-901-5888). Furthermore, 2θ values of 29.15° , 31.43° , 36.03° , 50.62° and 60.16° indicate orthorhombic CuO_2 with the (200), (400), (111), (131) and (440) planes, respectively (entry # 96-152-1321). Lattice parameters of CuO_2 are given as $a = 11.4 \text{ \AA}$, $b = 7.30 \text{ \AA}$, $c = 2.73 \text{ \AA}$ and $Z = 4$ with the space group $F222(22)$. The rest of the labelled peaks reveal the ITO because of the use of ITO as the substrate. Previous studies indicate that the formation of copper oxide phases depended on annealing temperature, time and amount of injected O_2 . For instance, Dolai et al. [14] reported that XRD pattern of a thin film deposited via reactive magnetron sputtering of Cu target with the Ar and O_2 ratio of 60:40 at 573 K revealed metallic Cu peaks and by increasing the temperature to 623 K only CuO

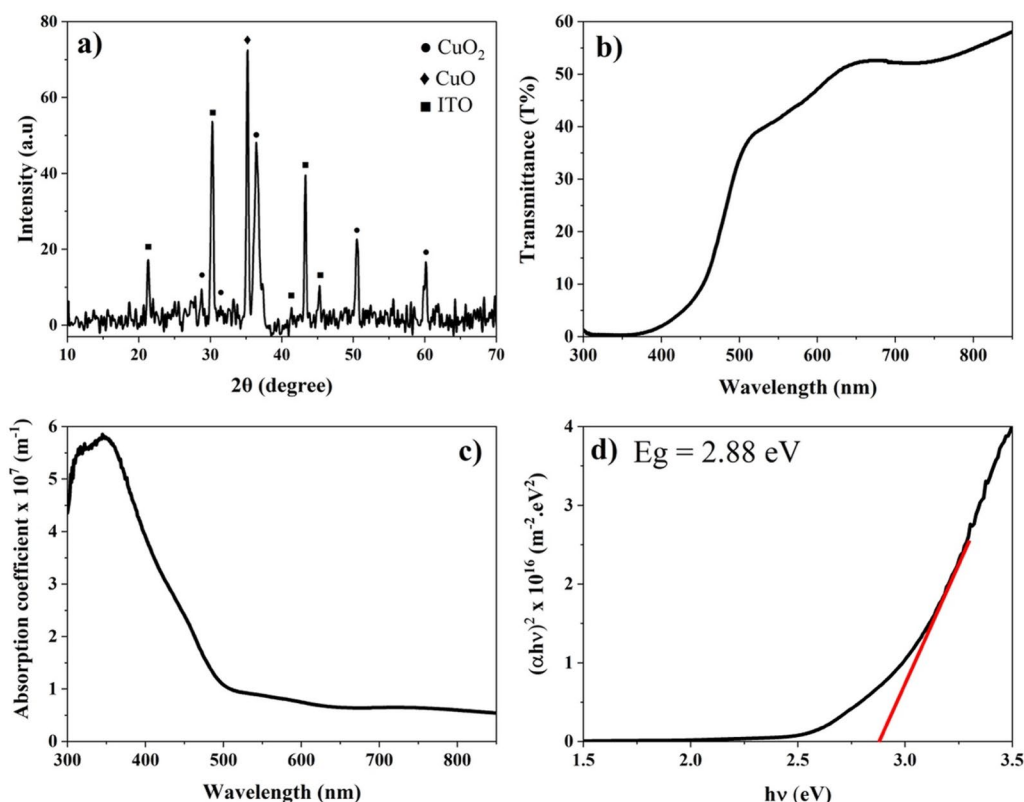


Figure 1: (a) XRD pattern, (b) transmittance, (c) absorption coefficient, (d) energy band gap for post-annealed CuO_x thin film deposited on ITO/glass substrate.

peaks exist. However, the XRD measurement reveals that two copper oxide phases coexist formed in the film and metallic Cu was not obtained. Hence, the XRD result indicates that magnetron sputtered copper film fully oxidised under oxygen environment via post-deposition annealing process at 473 K under high vacuum conditions and crystallised as two copper oxide phases.

Figure 1(b) illustrates the transmittance versus wavelength plot of CuO_x thin film deposited on ITO/glass substrate. Transmittance was determined via its relationships with absorbance ($A = -\log T$) with the formula of $T = (1 - R)^2 \exp(-A)$ which neglects the multiple reflections [21]. The transmittance values of the CuO_x thin film in the visible region initially sharply increased from 2.37 to 38.1% with increasing the wavelength from 400 to 530 nm, thereafter, it reached to 53.76% at 780 nm. Figure 1(c) displays the absorption coefficient of the deposited CuO_x thin film, determined via the equation of $\alpha = 2.303 A/d$, where A and d refer to the absorbance and the thickness of the film [22], respectively, versus wavelength. The strongest absorption peak of CuO_x thin film was obtained at 370 nm with the value of $5.8 \times 10^7 \text{ m}^{-1}$. As illustrated in Fig. 1(d), the energy band gap (E_g) value of the CuO_x thin film was identified by extrapolating the linear portion of $(\alpha h\nu)^2$ to the photon energy ($h\nu$) axis. Figure 2(d) was extracted from the Tauc plot method [23] which utilises the absorption coefficient (α)-related equation, $\alpha(h\nu) \approx (h\nu - E)^n$, where n is 1/2 for direct allowed transition [24,

25]. It can be noted that energy band gap of the CuO_x thin film deposited on ITO/glass substrate was measured as 2.88 eV.

Morphology of the CuO_x thin film was evaluated from the top-view SEM micrographs, which were taken at 10 k, 20 k and 40 k magnifications and are shown in Fig. 2(a)–(c), respectively. Low-magnification image displayed in Fig. 2(a) shows that CuO_x thin film homogeneously formed on ITO/glass. It might be observed from high-magnification images that the thin film has rough surface with nanoscale particles formed on top of the film. While the diameter of the nanoscale particles was denser in the size range of 100–300 nm, upto 800 nm particles also exist as shown in Fig. 2(c). EDX-mapping image [shown in Fig. 2(d)] demonstrates that the deposited copper film homogeneously oxidised thanks to post-deposition annealing process and this supports the XRD result presented in Fig. 1(a). Additionally, the proportions of elemental weight of Cu and O were measured as 79.45 and 20.55, respectively.

Optical characterisation of $\text{H}_2\text{Pc}/\text{ITO}$ and $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunctions

Optical characterisation of fabricated heterojunctions is an important aspect to establish their compatibilities of optoelectronic applications. Hence, crucial optical parameters such as absorption coefficient (α), optical transmittance ($T\%$), reflectance ($R\%$), energy band gap (E_g), extinction coefficient (k),

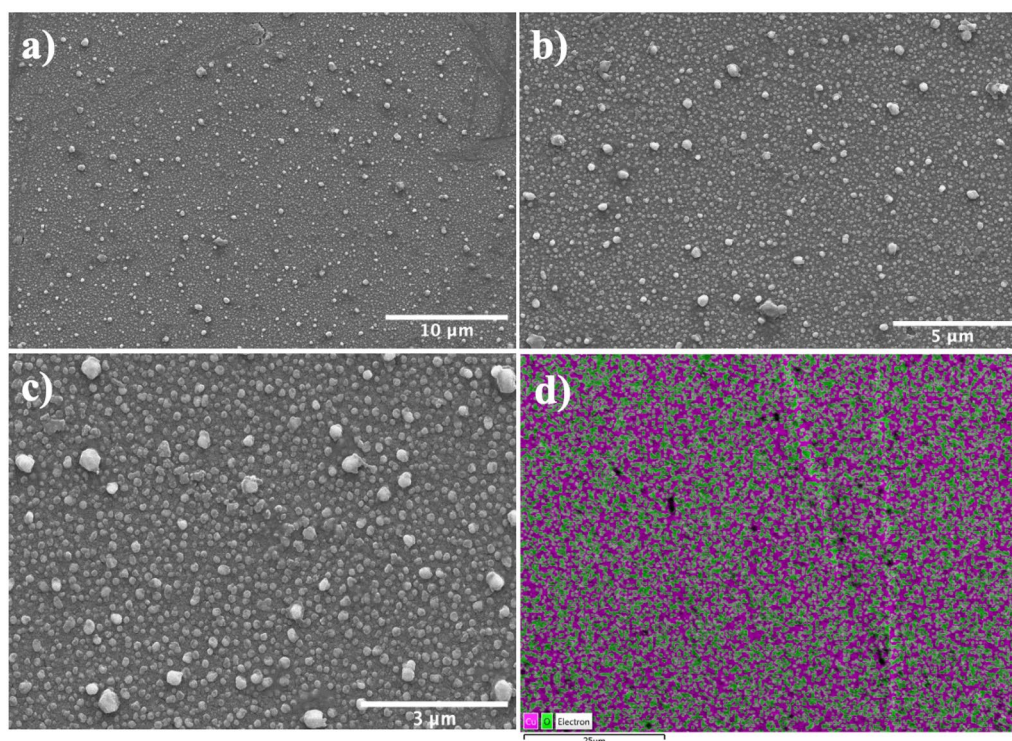


Figure 2: SEM micrograph of CuO_x thin film (a) at 10 k, (b) 20 k and (c) 40 k magnifications. (d) EDX-mapping image of CuO_x thin film (Purple and green indicate the copper and oxygen, respectively).

refractive index (n), optical conductivity (σ), real dielectric constant (ϵ_r and imaginary dielectric constants (ϵ_i), and dielectric loss (ϵ_i/ϵ_r) [26–29] of $\text{H}_2\text{Pc}/\text{ITO}$ and $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunctions were deeply evaluated between the 300 and 800 nm wavelength range.

The transmittance of $\text{H}_2\text{Pc}/\text{ITO}$ and $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunctions as a function of wavelength are illustrated in Fig. 3(a). While the transmittance of $\text{H}_2\text{Pc}/\text{ITO}$ was fluctuated between the values of 65% and 93% and reached its maximum value of 93% at 480 nm, that of $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ started to increase after 400 nm and reached its maximum value 38% at 600 nm. The big difference between the transmittance values is due to the CuO_x thin film which has 47% transmittance values at 600 nm as shown in Fig. 1(b). The strongest absorption coefficients of $\text{H}_2\text{Pc}/\text{ITO}$ and $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ were calculated as $2.2 \times 10^6 \text{ m}^{-1}$ at 390 nm and $1.7 \times 10^7 \text{ m}^{-1}$ at 340 nm, respectively [see Fig. 3(b)]. While absorption coefficients of $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunction sharply decreased to $4.2 \times 10^6 \text{ m}^{-1}$ till 510 nm and thereafter approximately stabilised, for $\text{H}_2\text{Pc}/\text{ITO}$ this was slightly fluctuated with increasing the wavelength. It can be clearly seen that there is an inverse correlation between $T\%$ and α in both the heterojunctions and this is because of the incident photons on the sample in the region with the lowest $T\%$ value have undergone maximum absorption. It is obvious that the $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunction had lower $T\%$ values and higher α values than the $\text{H}_2\text{Pc}/\text{ITO}$

heterojunction. This can be explained by the fact that the CuO_x layer plays a crucial role in photon absorption. Reflectance of both the heterojunctions was defined via the equation of $R = 1 - (T/\exp(-A))^{1/2}$ [30] as a function of wavelength and is demonstrated in Fig. 3(c). The reflectance value of $\text{H}_2\text{Pc}/\text{ITO}$ heterojunctions varied between 2.1 and 4.7% due to its high transmittance. Strongest reflectance of $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ was obtained at 310 nm with the value of 9.7% and it decreased to 6.7% at 800 nm. Furthermore, direct energy band gap values of the heterojunctions are displayed in Fig. 3(d) and they were determined as 2.66 eV and 2.77 eV for $\text{H}_2\text{Pc}/\text{ITO}$ and $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunctions, respectively. This difference of the direct energy band gap of the H_2Pc thin film is related to the substrate on which it is amplified. Organic-based thin films may exhibit different behaviours on different substrates due to their nature [5–7, 31].

Figure 3(e) and (f) shows the variation of extinction coefficient ($k = \frac{\lambda \cdot \alpha}{4\pi}$) and refractive index ($n = \frac{1+R}{1-R} + \sqrt{\frac{4R}{1-R^2} - k^2}$) for both the heterojunctions with wavelengths. It is seen that the k value of the $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunction increased with the increase of wavelength, reached a maximum value of 0.45 at 350 nm, then rapidly decreased and had a value of 0.17 at 500 nm and then slightly increased. Besides, k value of the $\text{H}_2\text{Pc}/\text{ITO}$ heterojunction fluctuated at short wavelength region and reached its maximum value of 0.09 at 800 nm by increasing slowly. While n values of $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunction exhibited generally

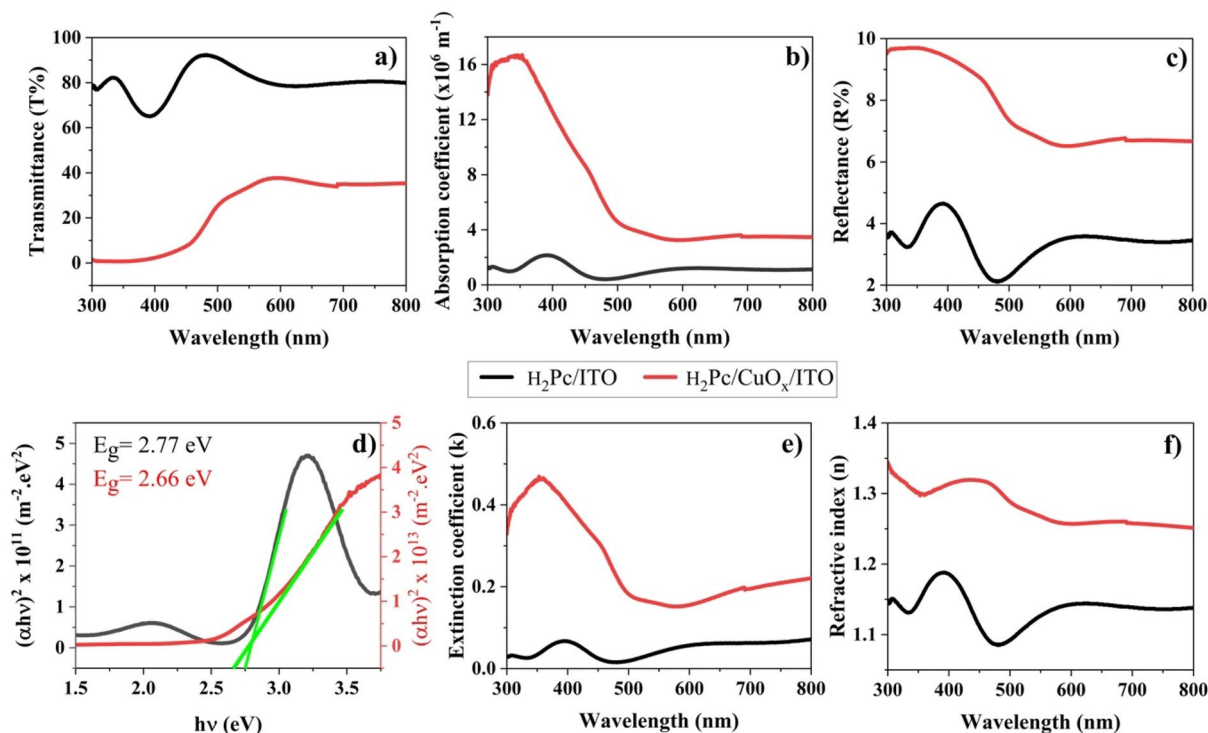


Figure 3: (a) Transmittance, (b) absorption coefficient, (c) reflectance of $\text{H}_2\text{Pc}/\text{ITO}$ and $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunctions as a function of wavelength. (d) $(\alpha h\nu)^2$ versus $h\nu$ plot for both the heterojunctions, (e) extinction coefficient (k) and (f) Refractive index (n) versus wavelength graphs of $\text{H}_2\text{Pc}/\text{ITO}$ and $\text{H}_2\text{Pc}/\text{CuO}_x/\text{ITO}$ heterojunctions.

decreasing behaviour with increasing the wavelength with an exception of small fluctuation around 430 nm, for H₂Pc/ITO heterojunction this peaked at 390 nm with the value of 1.19 and decreased to 1.09 at 430 nm and reached its beginning value with increasing the wavelength. It is also obvious that the H₂Pc/CuO_x/ITO heterojunction had higher k and n values than the H₂Pc/ITO heterojunction at all wavelengths and this became more pronounced especially in the absorption region. Generally, higher k and n values mean more absorption of light incident on heterojunctions and over-absorbed photons lead to overproduction of photogenerated carriers, which is very important in optoelectronic applications.

The optical conductivity ($\sigma = \alpha nc/4\pi$), imaginary dielectric constant ($\varepsilon_i = 2nk$), real dielectric constant ($\varepsilon_r = n^2 - k^2$) and dielectric loss ($\varepsilon_i/\varepsilon_r$) are the important parameters of the electronic excitation profile of the heterojunctions, and are given as a function of wavelength in Fig. 4. The highest values of σ , ε_r , ε_i and $\varepsilon_i/\varepsilon_r$ for H₂Pc/CuO_x/ITO heterojunction were obtained as 5.1×10^{15} at 360 nm, 1.65 at 480 nm, 1.2 at 360 nm and 0.9 at 370 nm, respectively. Furthermore, they were 7×10^{14} at 400 nm, 1.4 at 400 nm, 0.19 at 400 nm and 0.11 at 400 nm for H₂Pc/ITO heterojunction, respectively.

Electrical characterisation of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag heterojunctions

Traditional I-V characterisation is the most common used method to investigate the electrical properties of semiconductors. Current conduction mechanism in semiconductors is determined via I-V characteristics. Hence, electrical characterisations of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices were investigated in the potential range of ± 3 V under dark and 20, 40, 60, 80 and 100 mW.cm⁻² light intensities. Semilogarithmic I-V plots of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices are shown in Fig. 5(a) and (b), respectively. It might be noted that both the heterojunctions exhibited asymmetric behaviours at low voltage region and series resistance effect was observed for the heterojunctions towards the high voltage region. While series resistance effect in the Ag/H₂Pc/ITO/Ag device was observed at the voltage values greater than 0.46 V, this was monitored at voltage region higher than 0.3 V in the Ag/H₂Pc/CuO_x/Ag device. The variations of the rectification ratio ($RR = I_{\text{forward}}/I_{\text{reverse}}$) values of the Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices under dark and light intensities in accordance with the applied potentials are displayed in Fig. 5(c) and (d), respectively. It is clear that RR values for

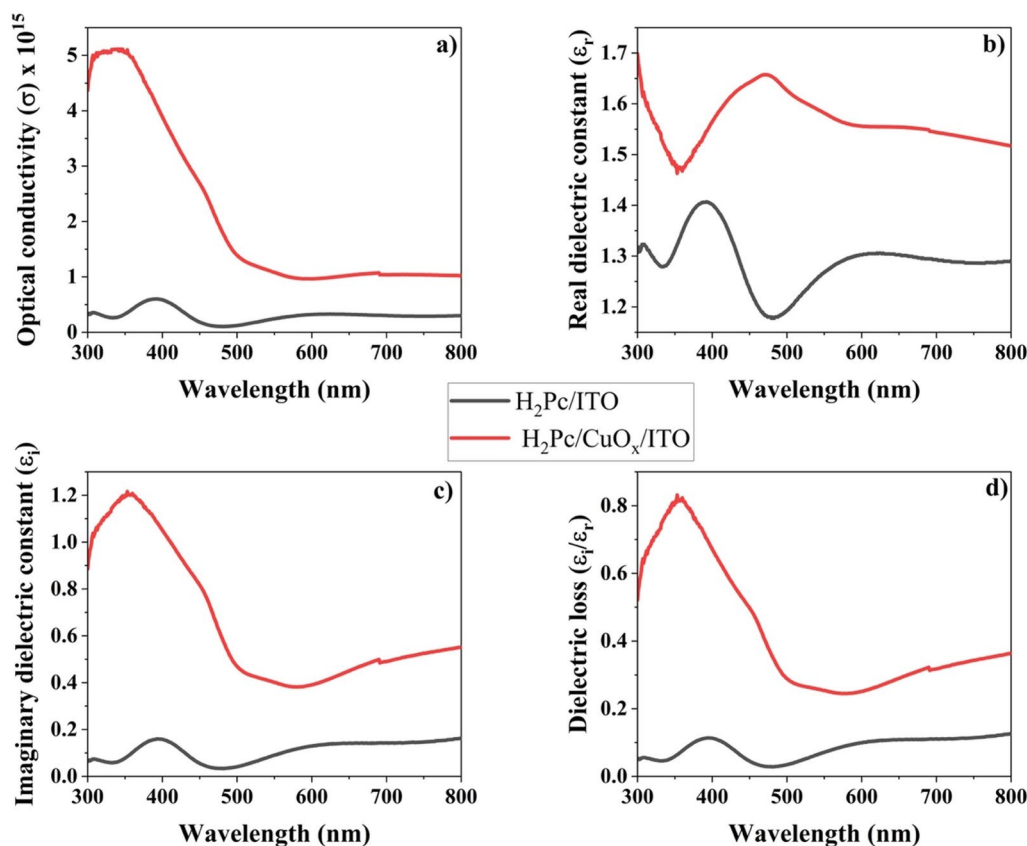


Figure 4: (a) Optical conductivity, (b) real dielectric constant, (c) imaginary dielectric constant and (d) dielectric loss of the heterojunctions as a function of wavelength.

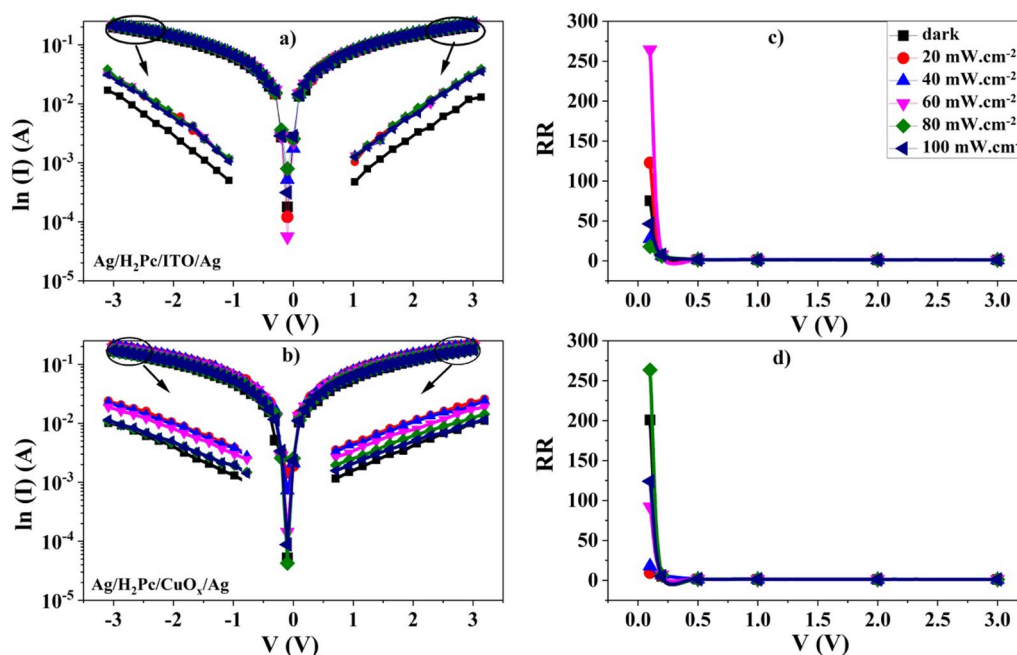


Figure 5: Semilogarithmic I - V plot of (a) Ag/H₂Pc/ITO/Ag (b) Ag/H₂Pc/CuO_x/Ag devices. The RR against V graph of (c) Ag/H₂Pc/ITO/Ag (b) Ag/H₂Pc/CuO_x/Ag heterojunctions.

both the devices exponentially decreased with the increase of applied voltage. While the RR values of Ag/H₂Pc/ITO/Ag device at 0.1 V under dark environment was about 75, this was 1.01 at 3 V. Similarly, while the RR values of Ag/H₂Pc/CuO_x/Ag device at 0.1 V under dark environment was approximately 200, this was 1.07 at 3 V. The decrease of RR values in accordance with the applied voltage can be explained with the charge accumulation at the interface and narrowing depletion region [32–34]. In addition, slow electron and hole mobility can be associated with the becoming of low RR value at high voltage values [35].

Electrical characterisation of I - V characteristic of heterojunctions exhibiting asymmetric behaviour can be analysed by thermionic emission theory [36] or by using the Cheung&Cheung functions [37] as well. Table 1 shows the variation of electrical parameters such as diode ideality factor (n), barrier height (ϕ_b), series resistance (R_s) and reverse saturation current (I_0) calculated by different methods, depending on light intensity. While the ideality factor of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices calculated by traditional method took values between 14.82 and 13.62 and 13.20 and 10.34, respectively, these calculated by Cheung&Cheung functions took values between 3.92 and 1.02 and 4.05 and 1.40. ϕ_b values of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices determined by the traditional method ranged between 0.428 and 0.424 eV, 0.436 and 0.430 eV, respectively. Besides, ϕ_b values of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO/Ag devices determined from Cheung&Cheung functions ranged between 0.09 and 0.34 eV and 0.11 and 0.30 eV, respectively. It can be stated that n and

ϕ_b values calculated by two different methods exhibited similar behaviours depending on light intensity, however, there were differences in the calculated values. These differences can be issued due to difference of the I - V regions used for calculations. Additionally, as a result of the increase of light intensity, while n values of both the heterojunctions decreased, ϕ_b values increased by exhibiting opposite behaviour which has been obtained in previous studies [38–40]. These changes of n and ϕ_b values under light intensity occurred in a fluctuating manner in both the heterojunctions as well and these could be attributed to the generation/separation of electron-hole pairs and reordering and restructuring molecular structures on the surface [38, 41]. n values of both the devices are far from the ideal diode ($n=1$) [42] but it is possible to encounter many studies involving heterojunctions where n values are greater than 1 [5, 43, 44]. There are several reasons that lead to n values greater than 1, can be listed as the regeneration and coupling of electron-hole pairs, distribution of interfacial charges, expansion of depletion layer, series resistance and non-homogeneous distribution of layers [41, 42, 45, 46].

Figure 6(a) and (b) illustrates the variation of ϕ_b values calculated via Cheung&Cheung functions as a function of n values of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices. Also, the graphs of the ϕ_b values determined by traditional method against n values of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices are displayed in Fig. 6(c) and (d). The equations in which ϕ_b values calculated from Cheung&Cheung functions of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices depend on n are

TABLE 1: Diode parameters of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices.

Light intensity (mW cm ⁻²)	Cheung and Cheung functions				Traditional method			Ohm's law
	dV/dlnI-I		H(I)-I		ln(I)-V			$\frac{\partial V}{\partial I}$
	Ag/H ₂ Pc/ITO/Ag							
	<i>n</i>	<i>R_s</i> (Ω)	ϕ _b (eV)	<i>R_s</i> (Ω)	<i>n</i>	ϕ _b (eV)	<i>I</i> ₀ (A)	<i>R_s</i> (Ω)
Dark	3.92	16.74	0.09	8.19	14.62	0.420	1.02×10 ⁻³	14.88
20	3.10	13.59	0.11	7.41	14.06	0.425	1.15×10 ⁻³	13.16
40	2.33	13.83	0.14	7.81	14.27	0.424	1.20×10 ⁻³	13.13
60	2.50	13.87	0.14	7.47	14.82	0.424	1.20×10 ⁻³	13.16
80	1.02	15.04	0.34	7.71	13.69	0.425	1.15×10 ⁻³	13.12
100	2.41	14.56	0.14	7.75	14.07	0.424	1.19×10 ⁻³	13.26

Light intensity (mW cm ⁻²)	Cheung and Cheung functions				Traditional method			Ohm's law
	dV/dlnI-I		H(I)-I		ln(I)-V			$\frac{\partial V}{\partial I}$
	Ag/H ₂ Pc/CuO _x /Ag							
	<i>n</i>	<i>R_s</i> (Ω)	ϕ _b (eV)	<i>R_s</i> (Ω)	<i>n</i>	ϕ _b (eV)	<i>I</i> ₀ (A)	<i>R_s</i> (Ω)
Dark	4.05	17.65	0.11	9.46	11.24	0.436	7.52×10 ⁻³	16.85
20	2.27	16.21	0.17	7.28	11.20	0.430	9.52×10 ⁻³	14.03
40	1.83	14.74	0.21	7.21	10.88	0.430	9.41×10 ⁻³	14.22
60	3.89	13.67	0.11	7.19	13.20	0.429	1×10 ⁻²	14.96
80	3.58	15.29	0.12	7.89	10.36	0.435	7.82×10 ⁻³	15.94
100	1.40	17.63	0.30	8.81	10.34	0.436	7.67×10 ⁻³	16.7

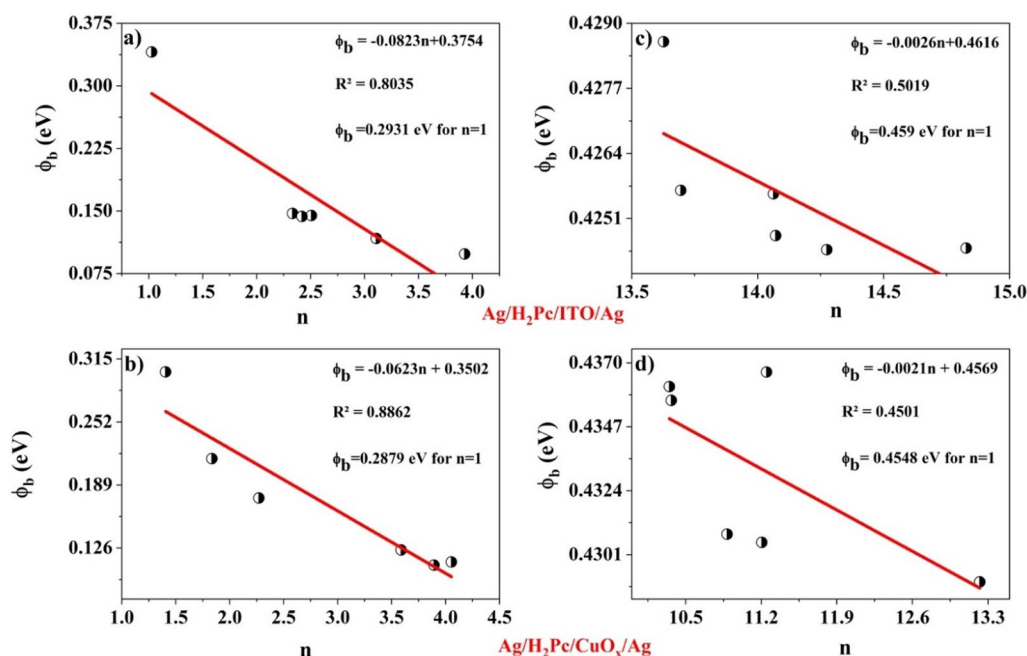


Figure 6: The graphs of the ϕ_b values calculated via Cheung& Cheung functions versus n values of (a) Ag/H₂Pc/ITO/Ag (b) Ag/H₂Pc/CuO_x/Ag devices. The graphs of the ϕ_b values calculated by the traditional method against to n values of (c) Ag/H₂Pc/ITO/Ag d) Ag/H₂Pc/CuO_x/Ag devices.

$\phi_b = -0.0823n + 0.3754$ and $\phi_b = -0.0623n + 0.3502$ and for $n = 1$, ϕ_b values are 0.2931 eV and 0.2879 eV, respectively. Additionally, equations of ϕ_b value calculated by traditional method for Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices are

$\phi_b = -0.0026n + 0.4616$ and $\phi_b = -0.0021n + 0.4569$ and for $n = 1$, ϕ_b values are 0.4590 eV and 0.4501 eV, respectively. It can be seen that ϕ_b values linearly depend on n values as illustrated in Fig. 6. Lowest and highest ϕ_b value for $n = 1$ was observed in

the Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices, respectively. It can also be stated that high ϕ_b values were obtained at low n values for both the heterojunctions. In general, charge carrier mobility increases with the increase of light intensity. Hence, due to the movement of these carriers, barrier height might increase and ideality factor might decrease. That is, as the light intensity increases, more electrons would have enough energy to overcome higher barrier [47]. Similar situations were studied by Monch et al. [48], Akkilic et al. [49] and Schmitsdorf et al. [50]. Additionally, the relationship between the high barrier height and low fidelity factor can be attributed as the bias dependence of the saddle point potential of an inhomogeneous barrier [49].

Series resistance (R_s) is a very crucial parameter for diodes and it significantly affects the device quality and performance [51]. The changes of R_s values depending on light intensity were determined by Cheung&Cheung functions and ohm's law ($R_j = \frac{\partial V}{\partial I}$). It is obtained that R_s values of fabricated devices were found to be quite low compared to the literature [40, 52–54]. R_s values decreased with increasing light intensity in both the devices and the lowest R_s value was observed in Ag/H₂Pc/ITO/Ag device. The decrease in R_s values can be explained with the increase in electrical conductivity under light [39]. Besides, I_0 values calculated by traditional method exhibited increasing tendency with the increase of light intensity. The minority carrier concentration also increased due to photon absorption, and similar situation has been observed by many researchers [55–57]. While I_0 of Ag/H₂Pc/ITO/Ag device took the values of 1.02×10^{-3} to 1.20×10^{-3} A, for Ag/H₂Pc/CuO_x/Ag device this obtained the values of 7.52×10^{-3} to 1.0×10^{-2} A.

Photocurrent values ($I_{ph} = I_{ill} - I_{dark}$) of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices calculated between the light intensity of 20 and 100 mW.cm⁻² at ± 3 V are given in Table 2. It is seen that I_{ph} values of Ag/H₂Pc/ITO/Ag device at -3 V and 3 V took the values between 1.66×10^{-3} and 2.12×10^{-3} A and 2.46×10^{-3} and 2.69×10^{-3} A, respectively. At the same time, for Ag/H₂Pc/CuO_x/Ag device, I_{ph} values at -3 V and 3 V were determined between 3.60×10^{-3} and 0.49×10^{-3} A and 3.58×10^{-3} and 0.16×10^{-3} A, respectively, and the alteration of I_{ph} as a function of light intensity at forward and reverse bias exhibited exponentially decreasing trend. Additionally, the highest I_{ph} values of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices were obtained at forward and reverse bias under 80 mW.cm⁻² and 20 mW.cm⁻² light intensities, respectively. The heterojunctions reached their maximum photogenerated and saturated at these light intensities. Compared to the I_{ph} values, it is obtained that Ag/H₂Pc/ITO/Ag device had higher values. The main reason for this can be attributed to the optical analyses showing that the H₂Pc/ITO heterojunction has higher α , k and lower E_g values than the H₂Pc/CuO_x heterojunction.

Photoresponsivity (R) is calculated from the ratio of the obtained photocurrent to incoming optical power with the equation of $R = I_{ph}/P_{inc}A$ [43], where P_{inc} is the light intensity and A (1.02×10^{-4} m²) is the area of the active layer. While R values of Ag/H₂Pc/ITO/Ag heterojunctions at -3 V and 3 V applied potential took the values of 0.81–0.15 A W⁻¹ and 1.28–0.24 A W⁻¹, respectively, for Ag/H₂Pc/CuO_x/Ag these were 1.76–0.03 A W⁻¹ and 1.75–0.01 A W⁻¹ as listed in Table 2. The highest R values of both the heterojunctions under forward and reverse bias were obtained at 20 mW.cm⁻¹ and they decreased with increasing the light intensity. The fact that the R values reach their highest value in a certain region and decrease in other light intensities can be explained by the decrease in absorption with the radial change of light intensity [58]. It may be noted that the obtained R values are higher than Co/CR/n-Si photodiodes [59] and p-Si/Fe(II)–polymeric complex/Au devices [60].

The percentage photosensitivity, $PS\%$, is determined via the equation of $PS\% = 100 \times (I_{ph} - I_d)/I_d$ [61–63] and I_d refers to the dark current. While the $PS\%$ values of Ag/H₂Pc/ITO/Ag heterojunction varied between 91.59 and 92.24% at -3 V and 113.38 and 112.22% at 3 V, for Ag/H₂Pc/CuO_x/Ag device these took values between 78.24 and 98.34% at -3 V and 120.10% and 100.88% at 3 V as shown in Table 2. The highest $PS\%$ value was obtained for Ag/H₂Pc/ITO/Ag device at 3 V under 20 mW.cm⁻² light intensity. This is also the region where the I_{ph} value is high. It is well known that oxygen plays an important role in the photosensitivity of diodes [64]. Thanks to the copper oxide layer, number of photogenerated electrons and holes significantly improved. This photoelectric advantage demonstrates that the optical transition works inside the nanolayers and supports an easy light collection procedure [65].

Specific detectivity (D^*) known as the detector capacity of weak light signals in photodiode is calculated via the equation of $D^* = R\sqrt{A}/(2qI_d)^{1/2}$ [66], where q refers to electron charge. D^* values were calculated for both the devices at ± 3 V potential range between 20 and 100 mW.cm⁻² light intensity and are listed in Table 2. While the calculated D^* values for Ag/H₂Pc/ITO/Ag device were found to be between 3.27×10^9 and 6.08×10^8 Jones at -3 V and 5.11×10^9 and 9.58×10^8 Jones at 3 V, for Ag/H₂Pc/CuO_x/Ag device this was obtained between 7.73×10^9 and 1.48×10^8 Jones at -3 V and 7.40×10^9 and 6.50×10^7 Jones at 3 V. It was obtained that D^* values decreased with the increase of the light intensity and the highest D^* values were found at 3 V under 20 mW.cm⁻² for both the devices. Similar behaviour was observed for I_{ph} and R . Compared to the literature, it was found that the calculated D^* values were lower than ZnO/NiO [67] and InSe/PMItz [31] heterojunctions.

Photovoltaic regions of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices are illustrated in Fig. 7 and photovoltaic parameters of the devices calculated between 20 and 100 mW.cm⁻² light intensities are given in Table 2. The calculation

TABLE 2: Photodetector and photovoltaic parameters of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices.

	Voltage (V)	Light intensity (mW cm ⁻²)	<i>I</i> _{ph} (A)	<i>R</i> (A W ⁻¹)	PS%	<i>D</i> [*] (Jones)
Ag/H ₂ Pc/ITO/Ag						
-3	20	1.66 × 10 ⁻³	0.81	91.59	3.27 × 10 ⁹	
	40	1.65 × 10 ⁻³	0.40	91.64	1.62 × 10 ⁹	
	60	1.64 × 10 ⁻³	0.27	91.68	1.08 × 10 ⁹	
	80	2.12 × 10 ⁻³	0.26	89.28	1.04 × 10 ⁹	
	100	1.53 × 10 ⁻³	0.15	92.24	6.03 × 10 ⁸	
3	20	2.62 × 10 ⁻³	1.28	113.03	5.11 × 10 ⁹	
	40	2.68 × 10 ⁻³	0.65	113.31	2.61 × 10 ⁹	
	60	2.62 × 10 ⁻³	0.43	113.04	1.7 × 10 ⁹	
	80	2.69 × 10 ⁻³	0.33	113.38	1.31 × 10 ⁹	
	100	2.46 × 10 ⁻³	0.24	112.22	9.58 × 10 ⁸	
Ag/H ₂ Pc/CuO _x /Ag						
-3	20	3.60 × 10 ⁻³	1.76	78.24	7.73 × 10 ⁹	
	40	3.32 × 10 ⁻³	0.81	79.96	3.56 × 10 ⁹	
	60	2.55 × 10 ⁻³	0.42	84.60	1.83 × 10 ⁹	
	80	0.27 × 10 ⁻³	0.03	98.34	1.48 × 10 ⁸	
	100	0.49 × 10 ⁻³	0.04	97.06	2.09 × 10 ⁸	
3	20	3.58 × 10 ⁻³	1.75	120.10	7.40 × 10 ⁹	
	40	3.29 × 10 ⁻³	0.80	118.46	3.40 × 10 ⁹	
	60	2.25 × 10 ⁻³	0.37	112.65	1.55 × 10 ⁹	
	80	1.01 × 10 ⁻³	0.12	105.68	5.24 × 10 ⁹	
	100	0.16 × 10 ⁻³	0.01	100.88	6.50 × 10 ⁷	
Light intensity (mW cm ⁻²)	<i>I</i> _{sc} (A)	<i>I</i> _{mpp} (A)	<i>V</i> _{oc} (V)	<i>V</i> _{mpp} (V)	FF%	η%
Ag/H ₂ Pc/ITO/Ag						
20	2.47 × 10 ⁻³	1.22 × 10 ⁻³	2.0 × 10 ⁻²	1.10 × 10 ⁻²	27.17	6.55 × 10 ⁻²
40	1.75 × 10 ⁻³	0.9 × 10 ⁻³	1.5 × 10 ⁻²	0.83 × 10 ⁻²	28.68	1.84 × 10 ⁻²
60	2.28 × 10 ⁻³	1.2 × 10 ⁻³	1.9 × 10 ⁻²	0.98 × 10 ⁻²	27.15	1.91 × 10 ⁻²
80	2.45 × 10 ⁻³	1.4 × 10 ⁻³	2.1 × 10 ⁻²	1.07 × 10 ⁻²	29.12	1.83 × 10 ⁻²
100	2.82 × 10 ⁻³	1.5 × 10 ⁻³	2.2 × 10 ⁻²	1.18 × 10 ⁻²	28.53	1.73 × 10 ⁻²
Ag/H ₂ Pc/CuO _x /Ag						
20	1.91 × 10 ⁻³	1.01 × 10 ⁻³	1.61 × 10 ⁻²	0.84 × 10 ⁻²	27.59	4.14 × 10 ⁻²
40	2.11 × 10 ⁻³	1.16 × 10 ⁻³	1.94 × 10 ⁻²	1.01 × 10 ⁻²	28.34	2.83 × 10 ⁻²
60	2.27 × 10 ⁻³	1.26 × 10 ⁻³	2.09 × 10 ⁻²	1.08 × 10 ⁻²	28.68	2.21 × 10 ⁻²
80	2.54 × 10 ⁻³	1.47 × 10 ⁻³	2.51 × 10 ⁻²	1.25 × 10 ⁻²	28.82	2.24 × 10 ⁻²
100	2.29 × 10 ⁻³	1.28 × 10 ⁻³	2.46 × 10 ⁻²	1.27 × 10 ⁻²	28.87	1.59 × 10 ⁻²

of the power conversion efficiency (η%) is performed as follows [68], $\eta\% = (V_{oc}I_{sc}/A.Pin).FF \times 100$, where *Pin* is the power of incident light per unit area and *FF* is the fill factor. Percentage fill factor, FF%, is calculated with the expression of $FF\% = (I_{mpp}V_{mpp}/I_{sc}V_{oc}) \times 100$, where *V*_{oc} is the open circuit voltage, *I*_{sc} is the short circuit current and *I*_{mpp} and *V*_{mpp} are the current and voltage at the maximum power point (mpp), respectively. *V*_{oc} values of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices took the values between 1.5 × 10⁻² and 2.2 × 10⁻² V and 1.6 × 10⁻² and 2.5 × 10⁻² V, respectively. Besides, the highest value of Ag/H₂Pc/ITO/Ag device was obtained at 100 mW.cm⁻² light intensity, while for Ag/H₂Pc/CuO_x/Ag device this was obtained at 80 mW.cm⁻² light intensity. It might

also be noted that increasing the light intensity led to increase of *V*_{oc} values of both the heterojunctions. In general, it is seen that the *V*_{oc} values of the Ag/H₂Pc/ITO/Ag device were higher than that of the Ag/H₂Pc/CuO_x/Ag device. The *I*_{sc} values of the Ag/H₂Pc/ITO/Ag device took values between 1.75 × 10⁻³ and 2.82 × 10⁻³ A with the highest value at 100 mW.cm⁻², while that of the Ag/H₂Pc/CuO_x/Ag device were between 1.91 × 10⁻³ and 2.54 × 10⁻³ A with the highest value at 80 mW.cm⁻². It is also seen that similar to *V*_{oc}, *I*_{sc} values increased with the increase of light intensity in both the devices, and this is typical photodiode behaviour. In addition, while the highest *FF*% value was 29.12% at 80 mW.cm⁻² light intensity Ag/H₂Pc/ITO/Ag device, it was measured as 28.87% at 100 mW.cm⁻² light intensity at Ag/

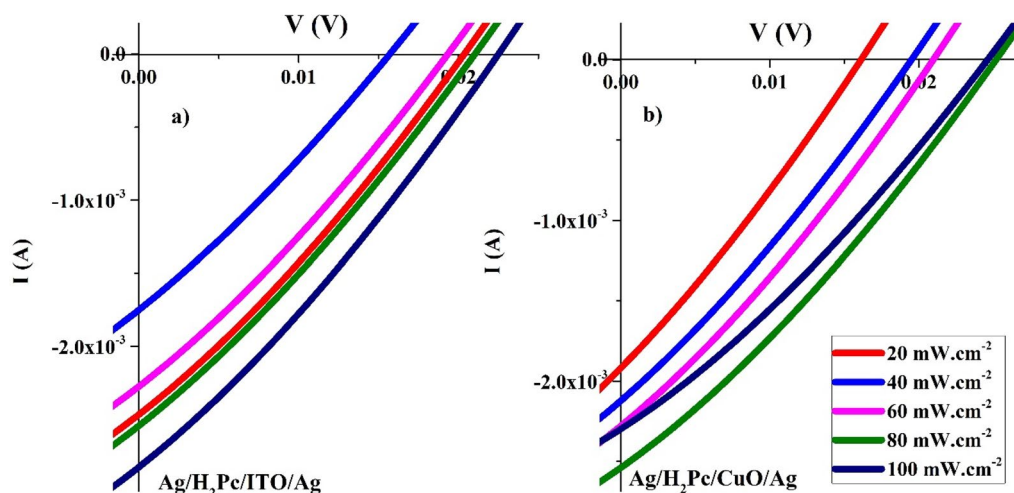


Figure 7: Photovoltaic regions of (a) Ag/H₂Pc/ITO/Ag (b) Ag/H₂Pc/CuO_x/Ag devices.

H₂Pc/CuO_x/Ag device. The $\eta\%$ values of the Ag/H₂Pc/ITO/Ag heterojunction were between 6.55×10^{-2} and 1.73×10^{-2} , while the $\eta\%$ values of the Ag/H₂Pc/CuO_x/Ag heterojunction were between 4.14×10^{-2} and 1.59×10^{-2} . In both the heterojunctions, the maximum $\eta\%$ value was observed at 20 mW.cm^{-2} light intensity. In general, $\eta\%$ values are found to be smaller compared with Pc-based devices in the literature [69]. It is seen that $\eta\%$ values decrease with increasing light intensity in both the devices. In general, the trend of the photodetector parameters against the light was also observed in the $\eta\%$ values. The radial change of light intensity with the increase of light intensity caused a decrease in $\eta\%$ values due to the decrease in absorption. At 20 mW.cm^{-2} light intensity, the maximum $\eta\%$ value of Ag/H₂Pc/ITO/Ag heterojunction is higher than Ag/H₂Pc/CuO_x/Ag heterojunction. This may be related to the high recombination between the CuO_x and the H₂Pc layer, as well as the high series resistance value. The fact that the V_{oc} values of the Ag/H₂Pc/ITO/Ag heterojunction are higher than that of the Ag/H₂Pc/CuO_x/Ag heterojunction is also the reason for high $\eta\%$ values. In addition, the reverse saturation current values of the Ag/H₂Pc/CuO_x/Ag heterojunction are higher than that of the Ag/H₂Pc/ITO/Ag heterojunction, which is the reason why the $\eta\%$ value is low. In addition to the high series resistance, the excess of leakage currents is one of the reasons for the decrease in efficiency in solar cells.

Conclusion

The modified H₂Pc organic semiconductors were successfully grown on ITO/glass and CuO_x/ITO/glass substrates. The energy band gap values of H₂Pc deposited on ITO/glass and CuO_x/ITO/glass were obtained as 2.77 and 2.66 eV, respectively. It is also obtained that optical parameters (α , R , k , n , σ , ϵ_i , ϵ_r and ϵ_i/ϵ_r)

of H₂Pc/CuO_x/ITO/glass heterojunction were quite higher than that of H₂Pc/ITO/glass heterojunctions. Electrical analyses of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices were performed under dark and several light intensities. It revealed that both the devices were sensitive to the light and exhibited rectifier characteristic especially in the low voltage region. Diode parameters (n , ϕ_b , R_s and I_0) of both the heterojunctions were determined by different methods (traditional method, Cheung&Cheung method and Ohm's law). In addition, the photodetector and photovoltaic parameters of both the devices were investigated at different light intensities. Consequently, it can be stated that both the devices having organic-based thin film technology might contribute novelty to the field of optoelectronic applications.

Materials and methods

The heterojunctions were fabricated on properly cleaned ITO/glass substrates with the following steps.

Synthesis of modified H₂Pc organic molecule

In the synthetic approach, firstly, compound 3 [see Fig. 8(a)] was synthesised by the SNAr-type reaction and then metal-free phthalocyanines are prepared from phthalonitriles in the presence of basic catalyst and synthesis and spectroscopic details of the compounds can be found in the literature [70].

Growing of CuO_x thin film

CuO_x thin film deposition on ITO/glass substrates was performed using a two-inch Cu (99.999% purity) magnetron target powered by the RF generator. Nanovak magnetron sputter system (Model:NVTS-400) was pumped down until it reached to 1×10^{-5} Torr base pressure. Deposition rate of Cu material

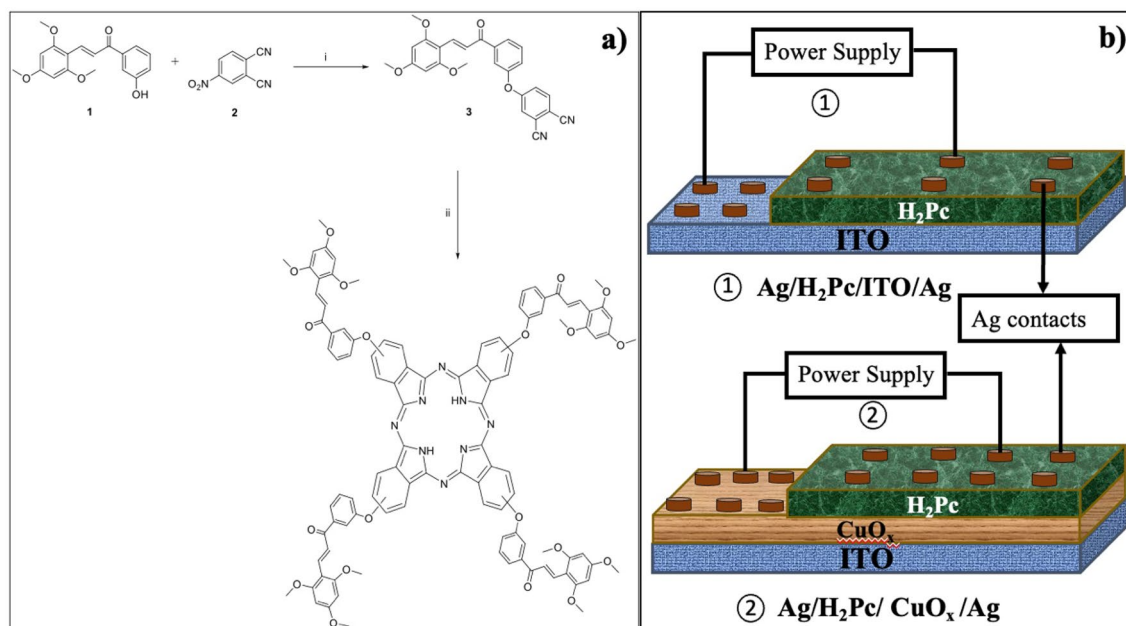


Figure 8: (a) Synthesis of modified H₂Pc organic molecule. (b) Schematic views of Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices.

was kept 1 Å/s by applying 65 W sputter power and 5 mTorr Ar gas injection under 3×10^{-3} Torr vacuum conditions. After reaching approximately 350 nm thickness, Cu film was annealed with 30 mTorr oxygen injection at 473 K during 30 min under high vacuum conditions.

Growing of modified H₂Pc thin film and fabrication of heterojunctions

The modified H₂Pc organic molecule was deposited on ITO/glass and CuO_x/ITO/glass surfaces as a thin film via the thermal evaporation method under high vacuum conditions. H₂Pc was placed in a molybdenum boat and evaporated controllably via the applied voltage at room temperature by employing Vaksis (PVD-MT/2M2T) thermal evaporator system. The working pressure of the deposition system was approximately 2×10^{-6} Torr. After ≈ 50 nm H₂Pc thin film was deposited on the ITO/glass and CuO_x/ITO/glass surface, a mask which has 0.77 mm diameter circular holes was applied to the top of the films. Thereafter, Ag contacts were grown on H₂Pc/ITO and H₂Pc/CuO_x surfaces via the thermal evaporation method, and Ag/H₂Pc/ITO/Ag and Ag/H₂Pc/CuO_x/Ag devices were fabricated as demonstrated in Fig. 8b.

Characterisation of the heterojunctions

Bruker D8 Advance X-ray diffractometer (XRD) was used to analyse the phase structure of CuO_x/ITO over a 10° to 70° range. The surface morphologies and the elemental

compositions of the CuO_x were determined via FEI Quanta FEG 250 Scanning Electron Microscope (SEM) equipped with Energy-dispersive X-ray spectrometry (EDX). Optical characterisation was performed via a Hach DR6000 model UV-vis spectrometer. Electrical characteristics of each heterojunction were examined under dark and illuminations between 20 mW.cm⁻² and 100 mW.cm⁻² via the FYTRONIX Solar Simulator LSS 900 *I*–*V* Characterisation System.

Author contributions

All authors contributed to the study conception and design. FU: drafting, coating of organic thin films and generation of heterojunctions, electrical analysis, conceptualisation, formal analysis and writing. SA: writing, editing, design and optical analysis. MSK: production of CuO_x thin film, writing and editing. MMK: *I*–*V* measurement, writing and editing. TA: synthesis of organic compound and writing. BC: *I*–*V* measurement and writing. MG: SEM and EDX measurement and writing. All authors read and approved the final manuscript.

Funding

There is no funding for this work.

Data availability

All data that support the findings of this study are included within the article (and any supplementary files).

Declarations

Conflict of interest The authors declare that they have no conflicts of interest.

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