



## Development of inorganic nanoparticle glycoconjugate enhanced cotton fabrics for multi-drug-resistant *Pseudomonas aeruginosa* bacterium

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### ARTICLE INFO

#### Keywords:

Antimicrobial cotton fabric  
Gold nanoparticle glycoconjugates  
Silver nanoparticle glycoconjugates and  
*Pseudomonas aeruginosa*

### ABSTRACT

Surface modification of cotton fabrics is required to advance their inherent properties in the development of antibacterial cotton fabric-based biomaterials. In this study, an antibacterial cotton fabric for multidrug resistant *Pseudomonas aeruginosa* species was developed. To do this, lactose and galactose derivatives were used to synthesize silver (Ag) and gold (Au) nanoparticle glycoconjugates (NP-GCs) to functionalize the cotton fabrics to target the carbohydrate binding proteins LecA and LecB lectins found on *P. aeruginosa*. The three lactose derivatives (i.e. Lactose sulfanilic acid (LSA), Lactose 5-aminosalicylic acid (L5AS) and Lactose 4-(4-aminophenyl) butyric acid (L4APB)) gave stable AgNP-GCs and AuNP-GCs while Galactose 4,4'-oxydianiline (GODA) allowed only stable AuNP-GCs synthesis. The cotton fabrics were pretreated to eliminate non-cellulosic parts to obtain scoured cotton fabrics (sCFs) for surface functionalization with AgNP-GCs and AuNP-GCs. Three different approaches (i.e. direct adsorption or adsorption through sulfhydryl group on sCFs and *in situ* synthesis) were followed to obtain AgNP-GCs functionalized cotton fabrics while AuNP-GCs functionalization was performed only through *in situ* synthesis. AgNP-GCs and AuNP-GCs were *in situ* synthesized on the sCFs under heat-treatment, and homogenous surfaces with high load of AgNP-GCs or AuNP-GCs were obtained. AuNP-GC-sCF were then functionalized with Colistin in order to add antibacterial property for *P. aeruginosa*. The colloidal AgNP-GCs, and *in situ* synthesized AgNP-GC-sCF and AuNP-GC-sCF showed strong antibacterial activity for *P. aeruginosa*. The formulations were then tested for gram (-) *Escherichia coli* and *Klebsiella pneumoniae*, and gram (+) *Staphylococcus epidermidis* to evaluate whether they have wide-spectrum antibacterial activity. While the colloidal AgNP-GCs showed similar high toxicity for these species in comparison to *P. aeruginosa*, only LSA\_AgNP-GC-sCF and LSA\_AgNP-GC-sCF showed 100 % growth suppression for *E. coli*, *K. pneumoniae* and *S. epidermidis*. The Colistin functionalized AuNP-GC-sCFs were also tested for Colistin resistant *K. pneumoniae*, and > 99 % growth suppression was obtained. We also tested whether the surface charge of the AuNP-GCs affect their interactions with *P. aeruginosa* using confocal laser scanning microscopy, where clear interactions were observed for GODA\_AuNPs. These early results revealed that inorganic nanoparticle glycoconjugates can be designed to develop antibacterial cotton fabrics through designing chemistry that can target lectins on bacterial membranes.

### 1. Introduction

Bacteria can gain resistance to conventional antibiotics, so today's strong antibiotics will be barely effective in the future, which will weaken the treatment efficiency (Burkatovskaya et al., 2008). It is

proposed that bacteria-specific nano-formulations, instead of conventional antibiotics, will be among the best possible options to fight against multi-drug resistant (MDR) bacterial infections (Wang et al., 2020). Among these nano-formulations, inorganic nanostructures (INs) are the main tools to develop antibiotic-free antibacterial agents against

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<https://doi.org/10.1016/j.microb.2025.100269>

Received 2 December 2024; Received in revised form 4 February 2025; Accepted 12 February 2025

Available online 13 February 2025

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MDR bacterial strains (Wounds and Nowak, 2021). In this perspective, silver nanoparticles (AgNPs) are among the most widely studied INMs because of their strong antibacterial activity (Khan et al., 2018). The antibacterial activities of AgNPs are resulted from a variety of mechanisms including oxidative stress, interference to ATP production, cell wall damage, interaction with nuclear materials and ribosome (Yin et al., 2020). For example, Pareek et al. (2021) showed that AgNPs dysregulated type II fatty acids biosynthesis and caused oxidative stress in MDR *Klebsiella pneumonia* (Pareek et al., 2021). Similarly, AgNPs were shown to impair DNA and accompanying protein synthesis in *P. aeruginosa*, which elevates oxidative stress and biofilm formation (Zhang et al., 2020). Therefore, the antibacterial power of the AgNPs has been transferred to biomedical applications, for instance, nanosilver coated cotton fabrics were used to develop antibacterial surfaces against the antibiotic-resistant bacterial species (El-Shishtawy et al., 2011). Besides the AgNPs, gold nanoparticles (AuNPs) are also convenient INMs used for antibacterial agent development studies as support materials carrying conventional antibiotics. AuNPs provide feasible support for the antibiotics to advance their antibacterial capacity even against the MDR bacterial strains (Sarma et al., 2024), and they can reduce the required amounts to bring out bactericidal effect (Sarma et al., 2024). Both silver nanoparticles (Ullah et al., 2022a, 2022b) and antibiotic functionalized AuNPs can also be used to coat surfaces in order to obtain different biomedical products. In this context, cotton fabrics are attractive base materials (Ghosh, 2021; İpek and Ertekin, 2021) owing to their strong mechanical (Chen et al., 2023a) and water-vapor absorbing characteristics (Chen et al., 2023b) in the development of antibacterial cotton fabrics (Gokce et al., 2020; Li et al., 2023).

Cotton fabrics (CFs) are mainly composed of cellulose (95 % (Knežević et al., 2022)) along with such impurities including waxes, pectin and proteins as of non-cellulosic components (Chung et al., 2004). It is the most abundant cellulosic material used in the textile industry (Knežević et al., 2022; Dang et al., 2024). Cellulosic materials provide unique platforms for nanomaterial introduction in a variety of applications (Gulati et al., 2022; Khan et al., 2017), including antibacterial fabrics (Raw and Fiber, 2023; Soleimani-Gorgani et al., 2023), UV protective fabrics (Nosheen et al., 2022), inflammable fabrics (Attia et al., 2023) and face-masks (Mast et al., 2023). Functionalization of CFs to advance the antimicrobial properties are mainly performed by introduction of nanomaterials and bioactive compounds (Shehabeldine et al., 2022). For example, silver/zinc doped cotton/paraffin (Wang et al., 2024) and amoxicillin loaded cellulose based composite (Qi et al., 2024) wound dressing fabric brought out broad antibacterial activity while contributing wound healing. The improved antimicrobial activity relies on a variety of mechanisms including controlled release of biocides, repulsion of microorganisms (e.g. via controlled hydrophobicity, electrostatic repulsion, alteration in surface topography and energy etc.) and capture of microorganism (for details (Muhammad et al., 2024)). For example, chitosan-zinc nanostructure introduced CFs possessed strong antibacterial capacity against both gram (+) and gram (-) bacterial species (Abdelhady, 2012). Similarly, copper oxide nanostructures (Nanoparticles, 2020; Wang et al., 2023), silver oxide nanoparticles and titanium oxide nanoparticles (Gao et al., 2021), titanium doped silver nanoparticles (El-Naggar et al., 2022) silver nanostructures (Raw and Fiber, 2023; Soleimani-Gorgani et al., 2023) were shown to be excellent nanomaterials advance the antimicrobial capability of CFs. Despite of the fact that successful applications of impregnation of pre-synthesized nanostructures onto cellulose based fabrics (Hu et al., 2024), *in situ* synthesis of AgNPs (Wu et al., 2024) and AuNPs (Anwar et al., 2021) onto the cotton fabrics to obtain functional and antimicrobial biomedical materials in order to provide a homogenous surface.

*Pseudomonas aeruginosa* is a motile rod-shaped biofilm-forming gram (-) and non-sporous bacterium that causes opportunistic infections (Pina-Sánchez et al., 2023). The life-threatening infections of *P. aeruginosa* are mostly related to such diseases as pulmonary infections, urinary tract infections, burn wounds, and cystic fibrosis

(Belgin and Öz, 2017). Since it develops resistance to antibiotics, the treatment and choice of antibiotics are not trivial (Pina-Sánchez et al., 2023). Particularly, MDR *P. aeruginosa* is listed as the greatest problematic bacterial species (Azam and Khan, 2019). Targeting LecA/LecB lectins are among the current approaches to eliminate MDR *P. aeruginosa* infection. Lectins are specific proteins that recognize free and/or bound carbohydrate residues on/in cells (Behzadi et al., 2021; Hu et al., 2019). LecA (Yu et al., 2019) and LecB (Sommer et al., 2014) selectively binds to lactose, galactose and fucose residues (Hu et al., 2019), and they have medicinal interest to prevent *P. aeruginosa* attachment to lung cells (Pertici and Pieters, 2012; Loris et al., 2003). However, to the best of our knowledge, this knowledge has not been transferred to the development of antimicrobial cotton fabric relying on carbohydrate-lectin interactions. In this study, we synthesized three lactose and one galactose derivatives that were used in the synthesis of silver and gold nanoparticle glycoconjugates, which were then introduced to scoured cotton fabric or used *in situ* synthesis of AgNP-GCs-sCF and AuNP-GCs-sCF. Due to the fact that AgNPs are potent antibacterial agents (Yaman Turan et al., 2024), AgNP-GCs-sCFs were used as synthesized while the antibiotic Colistin (targeting gram negative bacterial species (Elias et al., 2021)) was used to functionalize AuNP-GCs-sCF to have antibacterial surfaces. The carbohydrate ligands served as recognition elements to capture bacterial species (Kumar et al., 2019), which improves the contact of AgNPs and Colistin to the bacterial cells. The developed materials were characterized by spectroscopic, microscopic and microbiological techniques. The findings showed that the AgNP-enhanced cotton fabrics and Colistin-modified AuNPs-enhanced cotton fabrics gave excellent antibacterial activity against *P. aeruginosa* and the other tested other gram (-) and gram (+) bacterial species.

## 2. Material and methods

### 2.1. Materials

Raw cotton fabric and viscous bleach (active chlorine content <5%) were purchased from a local store. Mueller Hinton Broth, sulfanilic acid (99 %), lactose monohydrate (microbiology grade), 5-aminosalicylic acid (99 %), 4-(4-aminophenyl) butyric acid (95%), Galactose (≥99 %), 4,4'-oxydianiline (97 %), borane dimethylamine complex (97 %), Mueller Hinton Agar, AgNO<sub>3</sub> (≥99.0 %), anhydrous acetone, 200-proof ethanol (≥99.5 %, ACS reagent) and sodium hydroxide pellets were purchased from Sigma-Aldrich. 18.2 MΩ pure water was obtained using the Humana Zeneer water purification system in our labs.

### 2.2. Scouring cotton fabrics

In order to prevent possible damage to the fibers of the cotton fabric during the scouring, mild conditions were used as described in the literature (Yiğit et al., 2021). 2:98 (v/v) commercial viscous bleach was applied to 400 cm<sup>2</sup> cotton fabric at room temperature for overnight. The whitened cotton fabric was soaked into 2 % aqueous NaOH (w/v) for 2 h at 60 °C in order to eliminate non-cellulosic components (Rezaee et al., 2021). IR spectroscopy is a sensitive technique to monitor chemical alteration on surfaces (Chung et al., 2004), so the treated cotton fabrics were monitored using ATR-FTIR spectroscopy (Bruker Alpha II ATR-FTIR).

### 2.3. Synthesis and characterization of silver and gold nanoparticle glycoconjugates

In the first step, lectin databases were used to find out relevant publications revealing possible lactose and galactose derivatives that can be used to target *P. aeruginosa* (Supplementary Material Section 1). The sugar ligands were synthesized using reductive amination method as detailed in literature (Saladino et al., 2021; Jeong et al., 2007) (Supplementary Material Section 2). The synthesis of AgNPs was simply

performed by mixing the sugar ligands and silver nitrate in pure water for up to 6 h incubation at room temperature to obtain stable AgNPs. This approach allows fast and stable inorganic nanomaterial synthesis as shown previously by our group (Yazgan et al., 2020, 2021). It is also known that carbohydrates can also allow synthesis of stable AgNPs (Khan et al., 2018), where they can serve as reducing, capping and stabilizing agent. The molar ratios of the sugar ligands to silver nitrate were 7.23, 3.61, 1.82, 0.91 and 0.45, so the AgNPs were named AgNP7.23, AgNP3.61, AgNP1.82, AgNP0.91 and AgNP0.45. In the case of AuNPs, colloidal golds were only synthesized using L5AS and GODA sugar ligands with 7.23 molar ratio of the sugar ligand to auric acid. Selection of the sugar ligand/metal ion molar ratio was based on our previous studies (Yazgan et al., 2020; Sancak et al., 2023a). GODA sugar ligand did not provide any stable AgNPs under the tested conditions.

UV-vis spectroscopy (Multiskan SkyHigh Microplate Spectrophotometer, Thermofisher Scientific) at 350–900 nm range was used to characterize surface resonance peaks of the nanoparticles while high-resolution transmission electron microscopy (TEM) (FEI TALOS F200S TEM 200 kV, Thermofisher Scientific, service was purchased from BUMER-Bayburt University) was used to characterize the morphology and crystal structure. ATR-FTIR ( $4000\text{--}400\text{ cm}^{-1}$ ) (ALPHA II Compact FT-IR Spectrometer, Bruker) was used to characterize the sugar ligand-AgNPs interactions.

#### 2.4. Preparation and characterization of cotton fabric nanoparticle glycoconjugate composites

In the case of silver nanoparticle enhanced cotton fabric production three different approaches were followed (Fig. 1). (i) 5 mL of the selected AgNPs (1.26 mg/mL Ag content) were dropped on the scoured CFs (sCFs) or Cysteine modified sCFs (Cysteine introduction was performed based on the procedure given in the literature (Bayisa et al., 2023)) followed by keeping in an air-tight glass chamber at room temperature for 6 h or 2 h under 50 °C treatment. Right after the incubation, the coated CFs were vortexed in pure water twice to eliminate unbound

AgNPs. In the second approach (ii), the scoured cotton fabric silver and gold nanoparticle glycoconjugates were obtained through *in situ* AgNPs or AuNP synthesis on the scoured cotton fabrics. Briefly, the sugar ligand was added to the sCF and incubated for 10 min to allow sCF to fully absorb the sugar ligand. Then AgNO<sub>3</sub> (2 mg/mL) or HAuCl<sub>4</sub>·3(H<sub>2</sub>O) (0.8 mg/mL) dissolved in 5 mL was added to the sugar ligand containing sCF and incubated in a sealed glass contained at 50 °C for 6 h. The molar ratios were 0.91 and 3.61 for L5AS/Ag<sup>+</sup>, and LAPA/Ag<sup>+</sup> and LSA/Ag<sup>+</sup> while it was 7.23 for all of the AuNP-GC-sCFs. After the incubation, the sCF-AgNPs were rinsed twice with excess water. The coated CFs were then immersed in 70 % ethanol, followed by dried in a Biosafety Level 2 cabinet. The dried CF-AgNPs were exposed to 254 nm UV light for 30 min. The sterilized coated-CFs were characterized using leakage test (UV-vis spectroscopic test), total volume (BET), ATR-FTIR and SEM-EDX (FEI Quanta FEG 250) techniques, and microbiological tests. In the case of AuNP-enhanced cotton fabric preparation, only the third approach was followed with the same concentrations, for which SEM-EDX (FEI Quanta FEG 250) and microbiological tests were performed for the characterizations.

#### 2.5. Microbiological characterizations of the AgNPs and coated cotton fabrics

All the studied AgNPs were tested for their antibacterial activities as described in the literature using Nutrient Broth and Nutrient Agar (Saladino et al., 2021). As of the first step, AgNPs were tested for their toxicity against *P. aeruginosa* (gram -) using minimum inhibitory (MIC), minimum bactericidal (MBC), and biofilm formation tests. *Escherichia coli* (gram -), *Klebsiella pneumoniae* (gram -), *Staphylococcus epidermidis* (gram +) were used to test the selectivity of the AgNPs. It should be noted that the all gram negative bacterial species were resistant to Colistin while *S. epidermidis* was resistant to Pencillin G (MIC values were >10 µg/mL). MIC and MBC values were found using the serial dilution method (Sancak et al., 2023b) and the alterations in biofilm forming capacity of the bacterial species were tested at different AgNPs doses.

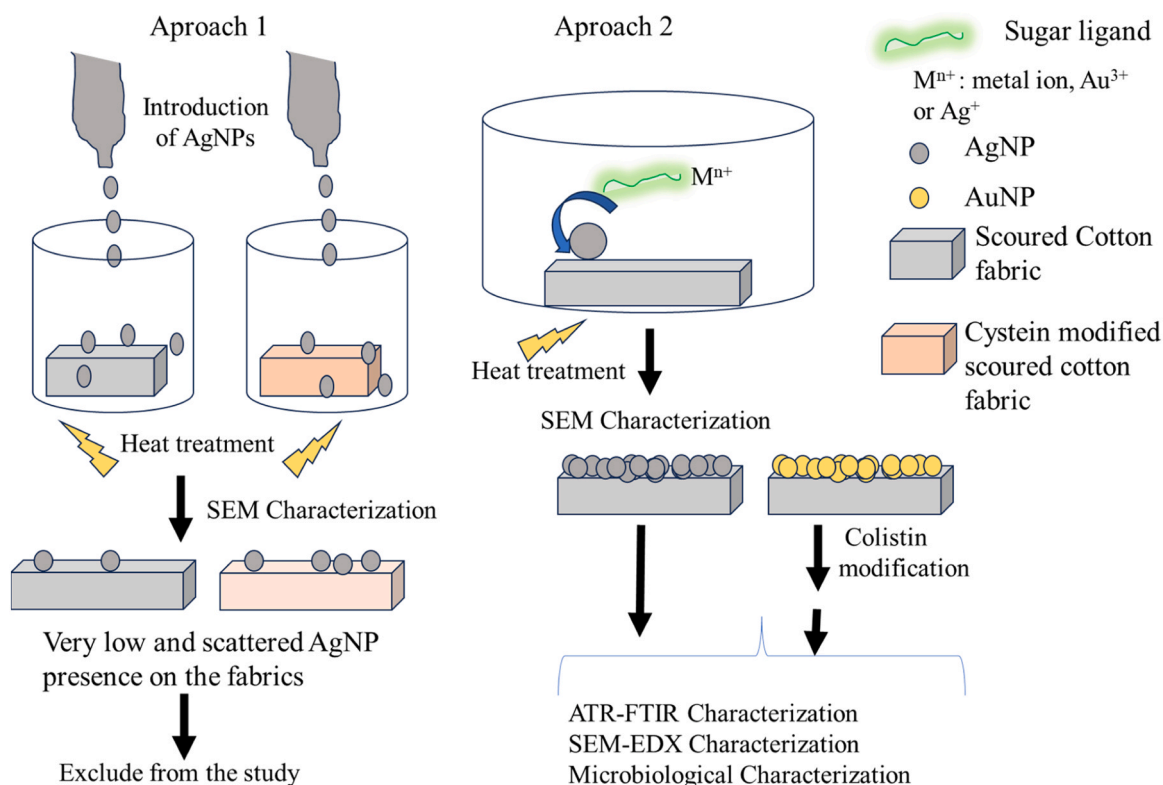


Fig. 1. Nanoparticle cotton fabric production approaches.

The antibacterial capacity of the AgNP-GC-sCFs and Colistin functionalized AuNPs-GC-CFs was tested by introducing them on freshly prepared agar, allowing them to submerge into the agar. This approach allows for monitoring if the bacterial cells pass through the pores of the developed fabric in comparison to the cotton fabric that was not coated with nanoparticles.  $10^6$  cfu/ $20 \mu\text{L}$  of the bacteria colonies were then introduced on the developed fabrics. Bacterial growth was checked right after an overnight incubation (16–18 h period). After the incubation, the cotton fabrics were placed into sterile tubes containing 5 mL of pH 7.4 PBS buffer and rinsed three times under 600 rpm vortexing. The developed cotton fabrics were then placed in 5 mL of broth and then incubated for 16 h. The turbidity of the samples was then read at 595 nm to monitor propagation of *P. aeruginosa* cells localized in/on the formulated cotton fabrics (Fig. 2).

The interaction between AuNP-GCs (i.e. GODA\_AuNP-GC and L5AS\_AuNP-GC, which are the two highly charged AuNPs) and *P. aeruginosa* was evaluated using confocal laser scanning microscopy (CLSM). For CLSM, bacteria *P. aeruginosa* ATCC 27853 from an exponentially growing culture in Luria-Bertani medium were washed twice by centrifugation (10 min, 5094 g; Sigma 3–16PK, Germany) with DI water and resuspended in DI water ( $\text{OD}_{600 \text{ nm}} = 0.5$ ;  $\sim 10^6$  cell forming units/mL) and exposed to the studied NPs at 5 mg/L for 1 h at room temperature. After the incubation, the cell suspension was stained by 5  $\mu\text{M}$  Syto9 (Invitrogen™ Thermo Fisher Scientific, USA) for 10 min in the dark. 20  $\mu\text{L}$  of the stained cell suspension was then spread on the microscopy slide, dried at 37 °C for 15 min and mounted in Mowiol (Sigma-Aldrich, Germany). The samples were visualized with Zeiss LSM800 CLSM (Zeiss, Jena, Germany) using a 488 nm laser and 505–550 nm emission filter to acquire a Syto9 signal. Nanoparticles were visualized by reflection mode at 640 nm. Images were processed using ZEN software (Carl Zeiss Microscopy, Jena, Germany).

### 3. Results and discussions

#### 3.1. Structural characterizations of the sugar ligands, nanoparticle glycoconjugates and nanoparticle glycoconjugate-cotton fabric formulations

##### 3.1.1. ESI-MS/MS characterization of the carbohydrate derivatives

ESI-MS/MS fragmentation pattern of LSA, L5AS, and L4APB were monitored using acetic acid/water (20:80, v/v) as the solvent, where protonation could add protons to the amino groups in addition to the proton came from positive screening mode. Therefore, we chose  $\pm 2 m/z$  during the screening in order to get reliable result (Yazgan, 2019). The fragment pattern showed that losses of 162  $m/z$ , 180  $m/z$  and 198  $m/z$  were common in LSA, L5AS and L4APB. The loss of 162  $m/z$ , 180  $m/z$  could refer to galactose moiety cleavage were in the form of pentose and

hexose moiety, respectively. The loss of 198  $m/z$  could be related to elimination of the water group right after the galactose moiety (as hexose) since it is common in all of them. In LSA and L5AS, the loss of 216  $m/z$  also refers to one more  $-\text{H}_2\text{O}$  elimination. Besides, the loss of 261.95  $m/z$  ( $180.05 + 81.08 m/z$ ) refers to the loss of the sulfonyl group accompanied by galactose moiety loss in LSA. GODA was also characterized using negative mode ESI-MS, and 526  $m/z$  was found (Supplementary Material Section 2).

##### 3.1.2. Infrared spectra analyses of the reactants

In the IR spectroscopic characterizations, the monomers and the possible residual contaminant sodium acetate should be analyzed. In the sodium acetate IR spectrum (Supplementary Material Section 3), the peaks at  $3407 \text{ cm}^{-1}$ ,  $3277 \text{ cm}^{-1}$ , and  $3165 \text{ cm}^{-1}$  are from the hydration of the carboxylate group with different hydration (Pitch et al., 2015) and protonation levels. The peak at  $2981 \text{ cm}^{-1}$  is from asymmetric  $\text{CH}_3$  stretching while the peak at  $1015 \text{ cm}^{-1}$  and  $1049 \text{ cm}^{-1}$  are from  $\text{CH}_3\text{-C}$  stretching and  $\text{CH}_3\text{-C-O}$  stretching accompanied with in-plane  $\text{CH}_3$  rocking, respectively (Kakihana and Nagumo, 1987). The peaks at  $1545 \text{ cm}^{-1}$  and  $1406 \text{ cm}^{-1}$  belong to asymmetric and symmetric COO, respectively (Kakihana and Nagumo, 1987; Habka et al., 2019) while the peaks at  $634 \text{ cm}^{-1}$  and  $510 \text{ cm}^{-1}$  are from COO bending and in-plane COO rocking (Kakihana and Nagumo, 1987). The two adjacent peaks at  $1696 \text{ cm}^{-1}$  and  $1635 \text{ cm}^{-1}$  are possibly related to the carbonyl functional group of carboxylate moiety (Susan Doofan et al., 2021). The peaks at  $925 \text{ cm}^{-1}$  and  $795 \text{ cm}^{-1}$  belong to COO-C stretching and  $\text{CH}_3$  rocking, respectively (Kakihana and Nagumo, 1987).

5-aminosalicylic acid (5AS) IR spectrum (Fig. 2) reveals the presence of the bands belonging to C-H stretching ( $2770 \text{ cm}^{-1}$ ), intramolecular hydrogen bonding with a carboxylic group ( $2492 \text{ cm}^{-1}$ ), C=O stretching of carbonyl group ( $1664$  and  $1572 \text{ cm}^{-1}$ ), N-H bending ( $1612 \text{ cm}^{-1}$ ), C=C stretching of aromatic group ( $1484 \text{ cm}^{-1}$ ), bending of -OH of carboxyl group ( $1444 \text{ cm}^{-1}$ ), Ar C-N stretching ( $1347 \text{ cm}^{-1}$ ), -OH stretching ( $1084 \text{ cm}^{-1}$ ) and C-H out of plane bending ( $808\text{--}706 \text{ cm}^{-1}$ ) (Hu et al., 2012; Singh et al., 2007).

Due to the fact that sugars give very strong IR peaks in the range of  $4000\text{--}400 \text{ cm}^{-1}$  and most of them are common in disaccharides and monosaccharides (Wiercigroch et al., 2017), we only focused on the characteristic glucose and galactose moieties to monitor the synthesis of lactose derivatives and their introduction to scoured cotton fabric in the form of sugar ligand-silver nanostructures (Supplementary Material Section 3). The following peaks belong to glucose unit of lactose, including  $3521 \text{ cm}^{-1}$  (stretching OH),  $1256 \text{ cm}^{-1}$  (rocking CH and OH),  $1166 \text{ cm}^{-1}$  (stretching CO),  $1064 \text{ cm}^{-1}$  (stretching CC),  $1028 \text{ cm}^{-1}$  (symmetric stretching CC),  $755 \text{ cm}^{-1}$  (twisting CO, COHO and HOH),  $547 \text{ cm}^{-1}$  (in-plane bending OCO),  $466 \text{ cm}^{-1}$  and  $439 \text{ cm}^{-1}$  (in-plane bending CCO). The characteristic galactose unit of lactose peaks are

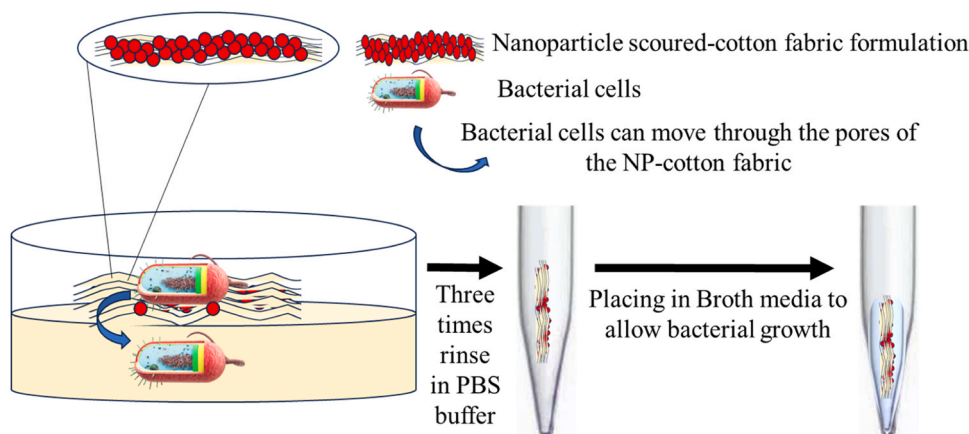


Fig. 2. Antibacterial activity testing flow of nanoparticle cotton fabric formulations.

1421  $\text{cm}^{-1}$  (wagging  $\text{CH}_2$ ), 1201  $\text{cm}^{-1}$  (in-plane bending OH and rocking  $\text{CH}_2$ ), 1103  $\text{cm}^{-1}$  (stretching CO), 673  $\text{cm}^{-1}$  (in-plane bending CCO) and 601  $\text{cm}^{-1}$  (in-plane bending OCO) (Wiercigroch et al., 2017).

Characteristic peaks belonging to 4ABP are 2997  $\text{cm}^{-1}$  O-H stretching, 2530  $\text{cm}^{-1}$  alkyl C-H vibration, 2385  $\text{cm}^{-1}$   $\text{CO}_2$  stretching, 1575  $\text{cm}^{-1}$   $\text{NH}_2^+$  deformation, 1418  $\text{cm}^{-1}$   $\text{CH}_2$  bending, 1374  $\text{cm}^{-1}$  C=O stretching, 1279  $\text{cm}^{-1}$  Ar C-N stretching, and 1203  $\text{cm}^{-1}$   $\text{CH}_2$  twisting vibration bands. The peak at 1203  $\text{cm}^{-1}$  belongs to  $\text{CH}_2$  twisting vibration while the peak at 1016  $\text{cm}^{-1}$  and 924  $\text{cm}^{-1}$  are from the aromatic ring and overlaps from the aromatic ring,  $\text{CH}_2$  and CC bonds, respectively. Besides, the wagging of  $\text{NH}_2$  group contributed to the peaks at 554 and 592  $\text{cm}^{-1}$ . However, a strong carbonyl band peak at  $\sim 1720$   $\text{cm}^{-1}$  and a medium  $\text{NH}_2$  deformation band peak between 3200 and 3400  $\text{cm}^{-1}$  range were not observed for this molecule because of the fact that inter- and intra-molecular hydrogen bonding occurred between carboxyl and amino group. Besides, amino peaks can couple with CC bond and CH vibrations of the phenyl ring, so the expected peaks become unclear (Akkaya et al., 2015) (Supplementary Material Section 3).

Sulfanilic acid undergoes dimerization, and its functional groups interact with each other even in solid form. Therefore, its IR spectrum might not provide all the expected groups when the IR analysis is performed at room temperature. Hereby, we assigned the characteristic peaks including O-H stretching band (2873  $\text{cm}^{-1}$ ), C-H stretching (2635  $\text{cm}^{-1}$ ),  $\text{NH}_2$  scissoring (1627  $\text{cm}^{-1}$ ),  $\text{NH}_3$  symmetric bending (1495  $\text{cm}^{-1}$ ), vibration from phenyl ring (1418  $\text{cm}^{-1}$  and 1004  $\text{cm}^{-1}$ ), C-N stretching (1239  $\text{cm}^{-1}$ ), in-plane bending of C-H and rocking  $\text{NH}_2$  of SA dimer (1111  $\text{cm}^{-1}$ ),  $\text{NH}_2$  rocking (1033  $\text{cm}^{-1}$ ), and  $\text{SO}_2$ -OH asymmetric bending of SA dimer (500  $\text{cm}^{-1}$ ) (Srimathi et al., 2014) (Fig. 2). Based on these assignments, it can be concluded that sulfanilic acid powder is in zwitterionic and dimer forms (Srivastav et al., 2023). Particularly, the peaks belonging to  $\text{NH}_2$  and Ar C-N bands are critical to monitor the successful substitution of sulfanilic acid to carbohydrates.

### 3.1.3. Infrared spectra analyses of the carbohydrate derivatives

In order to simplify the interpretation, the IR spectra of the sugar ligands are, here, discussed together (Fig. 3). It is noteworthy to mention that the sugar moiety tends to bend over the aromatic substituted group for these types of ligands (Yazgan et al., 2021), so the characteristic peaks of galactose moiety are not expected to give shifts at exactly the same wavenumber as they are seen in pure-lactose. Upon sugar ligand formation, a strong wide shift lying between 3660 and 3020  $\text{cm}^{-1}$  with a peak at  $3328 \pm 4$   $\text{cm}^{-1}$  belonging to -OH stretching concealed the three peaks of lactose at  $\sim 3521$   $\text{cm}^{-1}$ , 3324  $\text{cm}^{-1}$ , and 3241  $\text{cm}^{-1}$ . The peak at  $2909 \pm 11$   $\text{cm}^{-1}$  belongs to  $-\text{CH}_2$  vibration of lactose moiety of the ligands as well. The peaks at 890  $\text{cm}^{-1}$ , 896  $\text{cm}^{-1}$ , and 892  $\text{cm}^{-1}$  can be assigned to galactose moiety (Wiercigroch et al., 2017) of LSA, L5AS, and L4APB ligands, respectively. The peak at 1272  $\text{cm}^{-1}$  belongs to the rocking or bending vibration of CH and OH groups in the LSA spectrum while those appeared at 1249  $\text{cm}^{-1}$  and 1252  $\text{cm}^{-1}$  in L5AS and L4APB spectra, respectively. It is essential to bear in mind that certain IR peaks from self-dimerized sulfanilic acid seen in SA IR spectrum (Srivastav et al., 2023) will not be visible in LSA IR spectrum due to the fact lactose moiety alters SA-SA interactions through excess -OH groups along with that lactose moiety bends over the phenyl ring (Yazgan et al., 2021), so we paid effort to specific peaks related to aromatic ring, sulfonyl group and the secondary amine group of LSA. The peaks belonging to zwitterion and dimeric forms of SA along with  $\text{NH}_2$  rocking are not present in LSA while vibration from  $\text{SO}_3\text{H}$  (sulfur-oxy) at 1404  $\text{cm}^{-1}$  is present in LSA (Rana Tomar and Venkatesan Jayakumar, 2023). The peak at 1599  $\text{cm}^{-1}$  could be from  $\text{ArC}=\text{N}$  while the peak at 1519  $\text{cm}^{-1}$  belongs to vibration from the aromatic ring (Ismail et al., 2014) in LSA. However, the peak at 1599  $\text{cm}^{-1}$  is quite sharp, so it could be resulted from the aromatic ring as well. The peaks at 1156  $\text{cm}^{-1}$ , 1117  $\text{cm}^{-1}$ , 1067  $\text{cm}^{-1}$ , 831  $\text{cm}^{-1}$ , 700  $\text{cm}^{-1}$  and 557  $\text{cm}^{-1}$  belong to  $\text{ArC-N-C}$  stretching,  $\text{SO}_2$  symmetric stretching, S-O(H) mode vibration, aromatic ring vibration and  $\text{SO}_2$ -OH asymmetric bending, respectively. Besides,  $\text{CH}_2$  bending at 1466  $\text{cm}^{-1}$ , CH/OH rocking/bending vibration at 1272  $\text{cm}^{-1}$ , galactose ring torsion at 779  $\text{cm}^{-1}$  and galactose bending

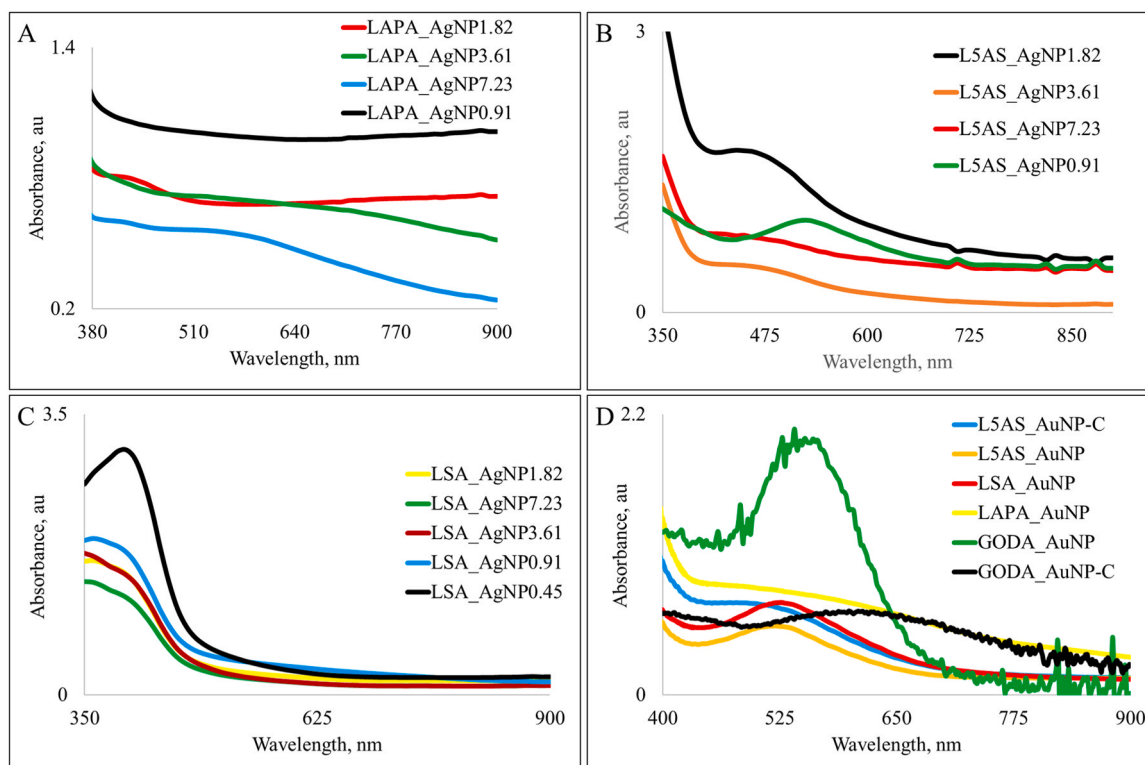


Fig. 3. UV-Vis spectra of LAPA (A), L5AS (B) LSA (C) synthesized colloidal AgNPs, and AuNPs and Colistin functionalized AuNPs (D).

vibration at  $626\text{ cm}^{-1}$  in the LSA spectrum belong to lactose residue. The galactose ring torsion peaks of L5AS and L4APB showed up at  $784$  and  $785\text{ cm}^{-1}$ , respectively. In the L5AS spectrum, carbonyl group vibration (e.g. stretching) showed up at  $1665$  and  $1626\text{ cm}^{-1}$  while -OH bending of the carboxyl group and ArC-N stretching bands showed up at  $1422\text{ cm}^{-1}$  and  $1319\text{ cm}^{-1}$ , respectively. The vibrations (e.g. C=C bending, C=C-H stretching) from the aromatic group of L5AS gave peaks at  $1493$ ,  $830$  and  $701\text{ cm}^{-1}$ . The carbonyl group gave vibration at  $1705\text{ cm}^{-1}$  in addition to the stretching carbonyl band at  $1392\text{ cm}^{-1}$  in L4APB. The peaks at  $1467\text{ cm}^{-1}$  and  $1142\text{ cm}^{-1}$  belong to N-H bending and ArC-N-C stretching, respectively. Aromatic ring vibration (e.g.  $1550\text{ cm}^{-1}$ ) and alkyl C-H vibration peaks appeared between  $810$  and  $432\text{ cm}^{-1}$  in L4ABPB. Through monitoring the peaks at  $925$  and  $795\text{ cm}^{-1}$  belonging to acetate, it was cleared that the sugar ligands did not contain acetate residue ([Supplementary Material Section 3](#)).

### 3.1.4. Infrared spectra analyses of the cotton fabrics

ATR-FTIR spectroscopy allows monitoring the changes on the surfaces, and successful applications were shown for the scoured cotton fabrics ([Chung et al., 2004](#)). Non-cellulosic components reside in different layers of cotton fabric ([Colombi et al., 2021](#)), we monitored entire IR spectrum to check whether any band belong to pectin and/or wax remained in the cotton fabric. The wide -OH stretching band lied between  $3620$  and  $3000\text{ cm}^{-1}$  in raw cotton fabric (rCF) shifted to  $3620$ – $3044\text{ cm}^{-1}$  range in the scoured cotton fabric (sCF). The -OH stretching band revealed its presence with two discernible peaks at  $3332$  and  $3271\text{ cm}^{-1}$  in sCF spectrum while there was not a discernible peak in rCF spectrum. Similarly, the -CH<sub>2</sub> vibration peaks at  $2914$  and  $2850\text{ cm}^{-1}$  in rCF shifted to  $2969$  and  $2897\text{ cm}^{-1}$  in sCF spectrum. The shift reveals that wax was removed from the cotton fabric since the peaks at  $2914$  and  $2850\text{ cm}^{-1}$  are from wax ([Chung et al., 2004](#)). There was no peak for the carbonyl group representing the carbonyl group from pectin in rCF and sCF spectra. However, sharper C-O-C, -OH, CH, and C-O vibration bands got relatively sharper and stronger, so it can be claimed that the scouring was successful (Figure was not shown).

IR spectra of the bleached and raw cotton fabrics were compared, and no difference was obtained. This is because raw cotton fabric is mostly formed from cellulose, and bleaching only removes such impurities as wax, residual proteins, and organic compounds. However, as it seen from [Supplementary Material Section 3](#), bleaching enhanced the IR bands upon removal of the impurities.  $3281\text{ cm}^{-1}$  band peak protonated OH stretching,  $2910\text{ cm}^{-1}$  C-H stretching,  $1427\text{ cm}^{-1}$  CH<sub>2</sub> wagging,  $1317\text{ cm}^{-1}$  C-H wagging,  $1158\text{ cm}^{-1}$  C-O-C pyranose ring and  $1023\text{ cm}^{-1}$  C-O-C stretching ([Nateghi and Hajimirzababa, 2014](#); [Sadnanand et al., 2017](#)). Physical adsorption of water molecules caused the formation of the broad band between  $3500$  and  $3100\text{ cm}^{-1}$  along with that of the peak at  $1610\text{ cm}^{-1}$  ([Igarashi et al., 2020](#)). The peaks at  $1105$ ,  $662$ , and  $602/502/436\text{ cm}^{-1}$  are related to  $\beta$ -glycosidic bond, H-bonded OH and CCC/COC glycosidic bonds, respectively ([Wiercigroch et al., 2017](#); [Nikonenko et al., 2005](#)).

The IR spectra of the sugar ligands did not show any dramatic changes upon silver nanoparticle (AgNP) formation ([Fig. 3](#)) and the corresponding sCF-AgNP ([Supplementary Material Section 3](#)). In all cases H-bonded -OH deformation peak at  $3666\text{ cm}^{-1}$  became clearer in the AgNPs, which appeared in the corresponding sCF-AgNP. The -OH stretching peaks at  $3332$  and  $3271\text{ cm}^{-1}$  in sCF became weaker upon AgNPs introduction while -CH<sub>2</sub> asymmetric and symmetric became stronger and sharper in the AgNPs and they became even stronger peaks than those obtained for -OH vibrations in sCF-AgNPs. The other clear difference was that the peaks at  $1052$  and  $1026\text{ cm}^{-1}$  in sCF spectrum turned into a stronger and shaper peak at  $\sim 1050\text{ cm}^{-1}$  in the sCF-AgNPs in all cases. However, there were not any other major changes, or any characteristic peaks were observed for LSA, L5AS or L4APB sugar ligands in the sCF-AgNPs. When similar studies were conducted for the sCF-AuNPs, no characteristic bands were observed for the sugar ligands as well (data not shown).

## 3.2. Characterizations of AgNP-GCs and AuNP-GCs

UV-Vis spectroscopic characteristics of the nanoparticles revealed that decrease in sugar ligand/Ag<sup>+</sup> ion ratio resulted in weakening of characteristic SPR peak for LAPA AgNPs; LAPA\_AgNP7.23, LAPA\_AgNP3.61 and LAPA\_AgNP1.82 gave SPR peaks at  $550$ ,  $530$  and  $440\text{ nm}$  while LAPA\_AgNP0.91 did not give any SPR peak, which could be related to at this ratio the AgNPs are not stable enough. In the case of L5AS ligand, L5AS\_AgNP7.23, L5AS\_AgNP3.61, L5AS\_AgNP1.82 and L5AS\_AgNP0.91 gave SPR peaks at  $470$ ,  $480$ ,  $460$  and  $530\text{ nm}$ , respectively. The first two SPR peaks are not sharp, which could be related to the fact that they are very small. In contrast to these, LSA synthesized AgNPs gave SPR peak at  $\sim 400\text{ nm}$  with an increasing intensity in response to decreasing LSA ligand/Ag<sup>+</sup> ion molar ratio. Colloidal AuNPs gave SPR peaks at  $523$ ,  $563$ ,  $542$  and  $570\text{ nm}$  for L5AS\_AuNP, LSA\_AuNP, GODA\_AuNP and LAPA\_AuNP, respectively. In order to show that Colistin can bind to AuNPs, GODA\_AuNPs and L5AS\_AuNPs were selected since they have the highest positive and negative zeta potentials; the zeta potentials were  $-34$ ,  $-24$ ,  $4$  and  $34\text{ mV}$  for L5AS\_AuNP, LAPA\_AuNP, LSA\_AuNP and GODA\_AuNP, respectively. GODA sugar ligand did not provide stable AgNPs under the tested conditions. Colistin functionalization ( $100\text{ ng Colistin}/1\text{ mg AuNP}$ ) altered the UV-Vis spectra pattern and SPR peak of GODA\_AuNPs (SPR peak shifted from  $596$  to  $632\text{ nm}$ ) and L5AS\_AuNPs (SPR peak shifted from  $523$  to  $520\text{ nm}$ ). Since Colistin was introduced to the media one hour after the nanoparticle synthesis started, it is possible that Colistin also affected the reaction kinetic and nanoparticles' morphologies, and it is known that Colistin can bind to both positively and negatively charged gold nanoparticles ([Fuller et al., 2020](#)).

HRTEM-SAED characterizations were performed for selected AgNPs in order to show how the sugar ligand/Ag<sup>+</sup> ion molar ratio and the ligand's chemistry affected morphology and size. L5AS\_AgNP0.91 nanostructures are anisotropic and generally pentagonal and minor hexagonal shaped with  $67\text{ nm}$  to  $234\text{ nm}$  rounded edges while mostly the edges are between  $140$  and  $170\text{ nm}$ . L5AS\_AgNP3.61 are a mixture of anisotropic 0D and 1D nanostructures ([Fig. 4](#)). Anisotropic AgNPs were sized between  $\sim 65\text{ nm}$  and  $132\text{ nm}$  while curved 1D structures were up to  $\sim 650\text{ nm}$  in length and  $112\text{ nm}$  in width.

Sizes between  $1.7$  and  $18\text{ nm}$  with overwhelmingly ellipsoid morphology, and an average of  $8.2\text{ nm}$ .  $\sim 50\%$   $5 \pm 1\text{ nm}$  range ([Fig. 5](#)). SAED analysis showed that the crystalline structure fits to *face centric cubic* lattice ([Yazgan et al., 2020](#)). The inset in [Fig. 7B](#) reveals that the nature of crystallinity has only one direction.

LSA\_AgNP3.61 gave no dominant size distribution with ellipsoid to irregularly shaped AgNPs while a minor portion of the AgNPs showed ellipsoidal morphology ([Fig. 6](#)). The general range of the AgNPs' edges ranged between  $\sim 20\text{ nm}$  and  $\sim 190\text{ nm}$ . In contrast to this, LSA\_AgNP0.45 gave mostly ellipsoidal-like morphology with edges between  $18$  and  $190\text{ nm}$ . Besides, some spherical and irregularly shaped anisotropic morphology were observed as well ([Fig. 6](#)). It is noteworthy that electron-beam triggered AgNPs were also formed within the grid (data not shown), which was not present for LSA\_AgNP2 sample. This shows that some of the Ag<sup>+</sup> ions did not turn into AgNPs, but interestingly they could not be separated from the mixture using centrifugation ( $4000\text{ g}$  for  $10\text{ min}$ ). This is probably related to the formation of large and continuous LSA-Ag<sup>+</sup> ion interaction.

Selected area diffraction (SAED) patterns of the AgNPs show that L5AS ([Fig. 4](#)) and LAPA ([Fig. 5](#)) sugar ligands directed face centric cubic lattice with  $(111)$ ,  $(200)$ ,  $(220)$ ,  $(311)$  and  $(31)$  miller indices (Reference code: 00-001-1167) ([Sancak et al., 2023b](#)). In contrast to these, LSA triggered formation of hexagonally packed crystalline lattice structure (Reference code: 01-087-0598) ([Fig. 6](#)), which can co-exist with *fcc* even though the crystal structure is overwhelmingly *fcc* ([Virgen-Ortiz et al., 2015](#)). Smaller LSA\_AgNPs were in the form of polycrystalline and larger ones were single crystalline. However, it should be noted that SAED pattern of LSA\_AgNP0.45 looked like crystalline phases

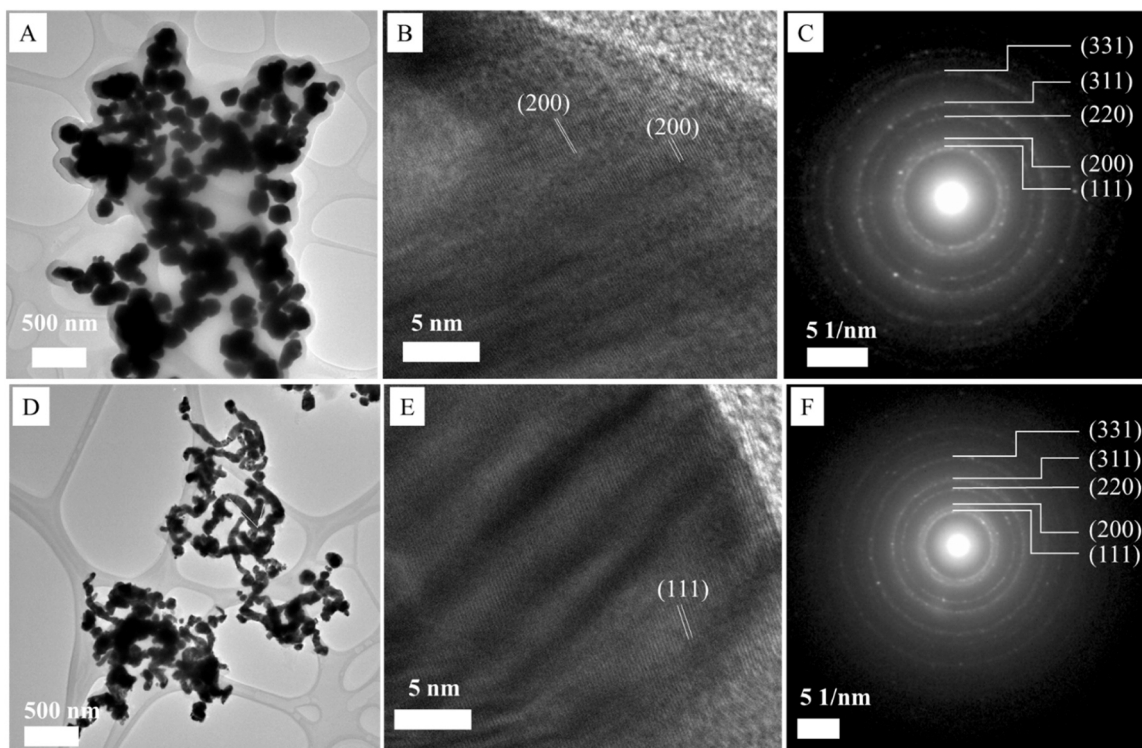


Fig. 4. TEM images of L5AS\_AgNP0.91 (A, B, D) and L5AS\_AgNP3.61 (C and F) and selected area diffraction of L5AS\_AgNP2 (E).

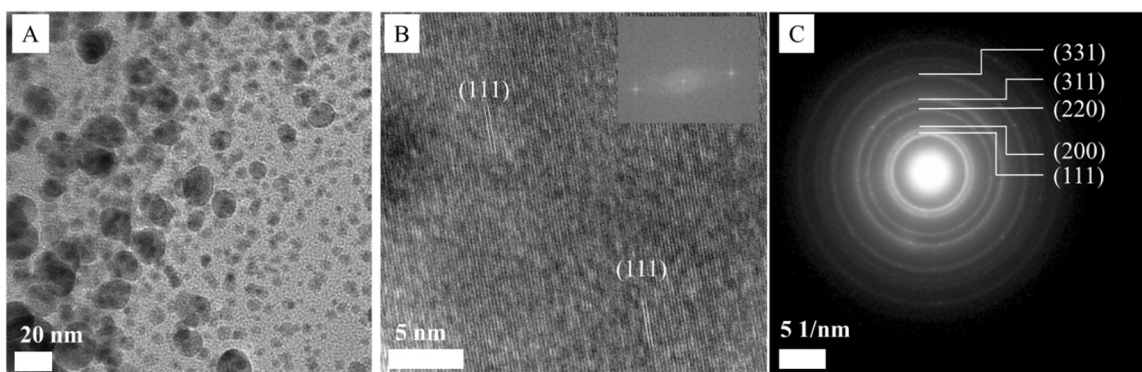


Fig. 5. TEM images (A/B) and SAED (C) of LAPA\_AgNP3.61.

overlapped each other. Besides, the sugar ligands might also undergo crystallization on the AgNPs since the interplanar distance (0.330 nm, Fig. 6B) did not meet with known interplanar distances of either face centric cubic crystalline structure or hexagonally packed crystalline structure. This is assigned to silver oxide nanoparticles, and it is known that silver oxide layers can form on AgNPs (Kumar et al., 2021). It is possible that the AgNP-GCs possess some silver oxide layers since there was heat treatment and same sugar ligands were used.

The average size of GODA\_AuNPs overwhelmingly fell into  $87 \pm 13$  nm range, but some reached up to 170 nm or down to  $\sim 35$  nm. GODA\_AuNPs were in the form of nanoclusters formed from smaller nanoparticles ( $\sim 9 \pm 3$  nm). At higher magnification, no crystal planes were observed, which can be attributed to low crystallinity of GODA\_AuNPs. In contrast to this, L5AS\_AuNP gave a very distinct morphology as spherical with a 9–12 nm range and highly crystalline (Fig. 7). The measured interplanar distances (between 0.232 and 0.24 nm) showed that L5AS\_AuNPs have *fcc* crystal structure as expected from the sugar ligand synthesized AuNPs (Sancak et al., 2023b). From the literature, SAED and XRD results show compatibility for the

sugar ligand synthesized gold and silver nanoparticles, so XRD studies were not performed.

Fig. 8 reveals *in situ* synthesized AgNP-GCs on the scoured cotton fabrics gave high-load and relatively homogenous presence of AgNP-GCs (plain sCF surface is quite smooth) while drop-casting of the AgNP-GCs on plain sCFs and Cysteine modified sCFs did not allow even distribution and high AgNP presence (Figure S6). L5AS\_AgNP-GCs on sCF gave 69.6 nm average sized AgNPs and mostly localized between 64 and 110 nm range with spherical to quasispherical morphology while LSA\_AgNP-GCs were overwhelmingly nanorod-like structures with 141.1 nm average size and mostly in 80–100 nm range while some reached up to 700 nm. LAPA\_AgNP-GCs were spherical, quasispherical, hexagonal and pentagonal with 97.5 nm average size (mostly localized in 60–110 nm range) (Fig. 8). In contrast to these, only L5AS\_AuNP-GCs on the sCFs gave dispersed AuNPs with 57.8 nm average size and mostly fell into 42–68 nm range. However, LSA\_AuNP-GCs, GODA\_AuNP-GCs and LAPA\_AuNP-GCs on the sCFs gave more film-like structures and covered the surfaces, for which 18–48 nm ranges,  $\sim 55$  nm and  $\sim 20$  nm were most common size ranges. Particularly, LSA\_AuNP-GCs and

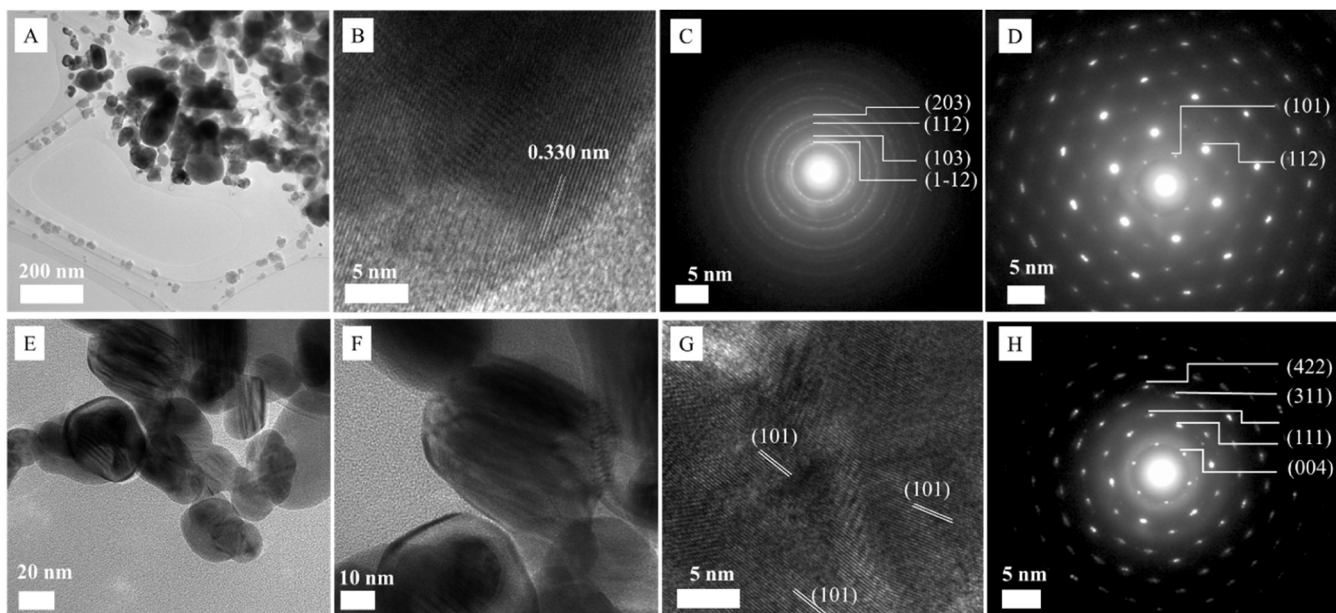


Fig. 6. TEM images and SAEDs of LSA\_AgNP3.61 (A/B and C/D) and LSA\_AgNP0.45 (E-G and H).

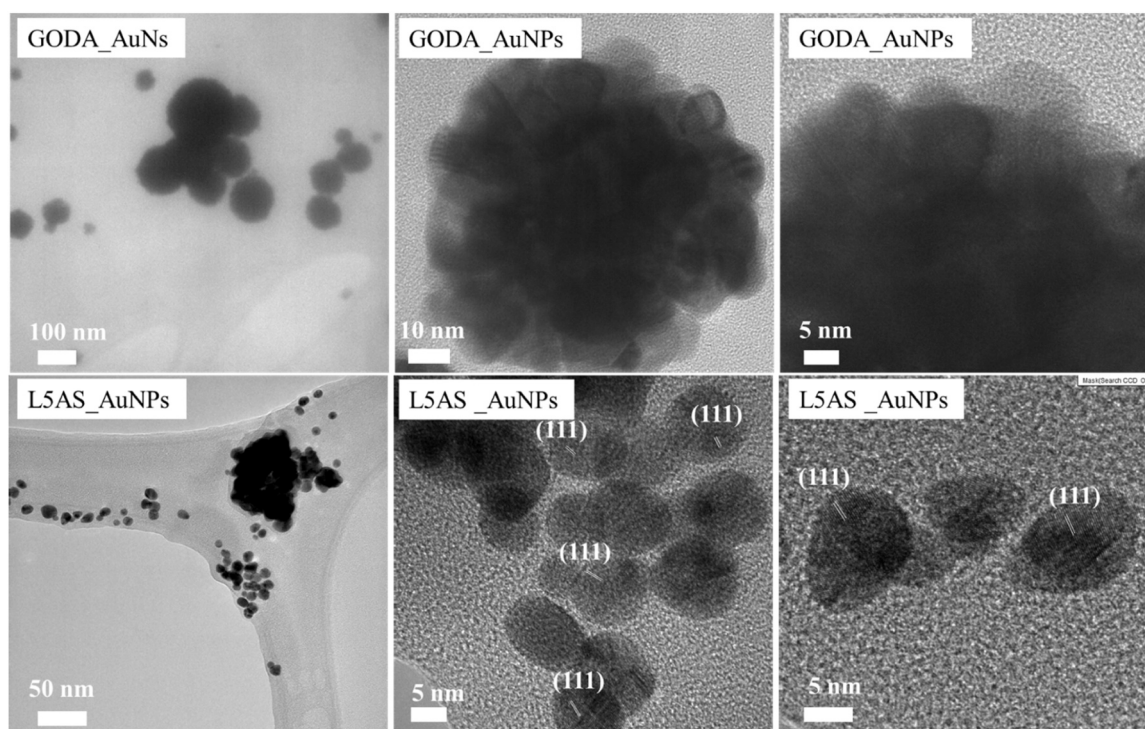


Fig. 7. TEM images of GODA\_AuNPs and L5AS\_AuNPs.

LAPA\_AgNP-GC seem to be that smaller nanoparticles agglomerate to form larger chunks on the fabric surface. In the light of literature (Zhu et al., 2021), AgNPs *in situ* synthesized on cotton fabrics do not agglomerate due to the presence of excess hydroxylic groups, which did not work for all the AuNPs. Therefore, it can be proposed that the carbohydrate ligands LSA and LAPA interactions with AuNPs were quite strong, so the hydroxylic groups could not show their capping character, which is critical in the prevention of agglomeration (Alamer and Beyari, 2022). For the AuNPs, donor-acceptor interactions are critical to prevent AuNPs agglomeration. Therefore, it is possible that the used LSA and LAPA ligand concentrations were higher than that of the optimum

concentration (Mehravani et al., 2021). Size and/or morphological differences between the colloidal and *in situ* synthesized AgNP-GCs and AuNP-GCs could be related to the heat treatment, or the cotton fabric altered the ligand-metal ion interaction. EDX analysis showed that the structures on the sCFs were silver (Figure S7) and gold (Figure S8), which revealed that the synthesis was successful. Besides, BET studies revealed that there was nearly no difference for the AgNP-sCFs and the control, which means that AgNPs mostly formed on the cotton fabrics, and they did not serve as fillers under the experimental conditions.

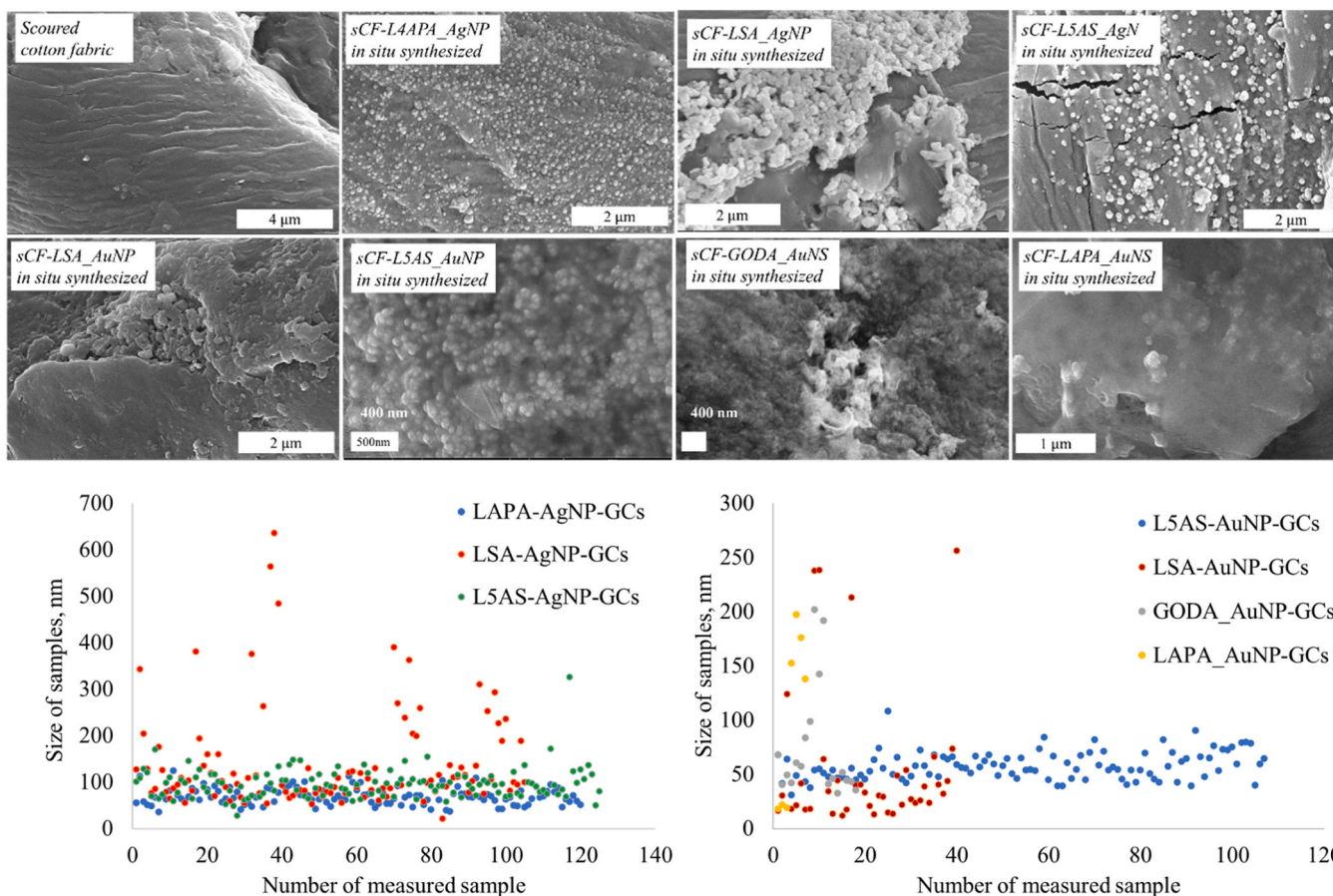


Fig. 8. SEM images of the scoured cotton fabric, nanoparticle-sCFs and their size distribution.

### 3.3. Antibacterial characterization

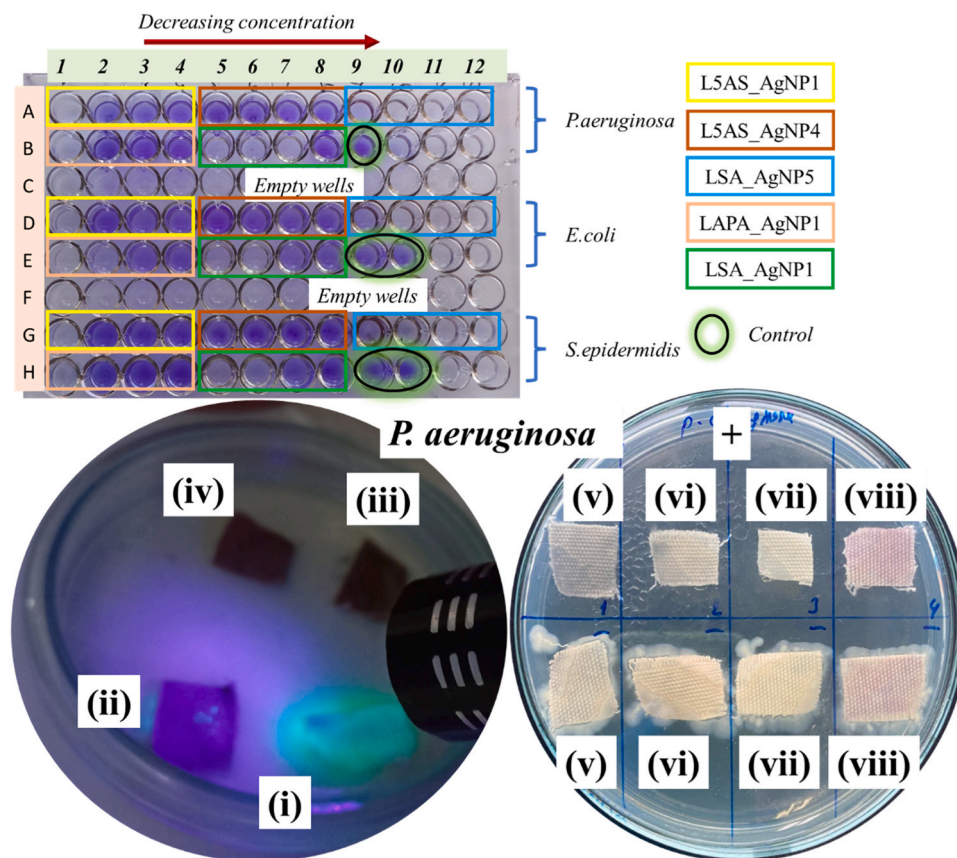
As seen from Fig. 9 and Table 1, L5AS\_AgNP1 and LAPA\_AgNP1 gave relatively lower antibacterial activity for the three bacterial species while L5AS\_AgNP4 did not give any antibacterial activity even at 1.26  $\mu\text{g}/\text{mL}$  concentration. LSA\_AgNP5 gave the highest antibacterial activity against all the tested bacterial species. Besides, LSA\_AgNP1 also showed strong antibacterial activity against *P. aeruginosa* and *E. coli* while it showed good antibacterial activity against *S. epidermidis* species. Similarly, the colloidal AgNP-GCs showed strong growth suppression and inhibition of biofilm formation for *K. pneumoniae* (Figure S9).

No bacterial growth was seen with bare-eye for the AgNP-GC-sCFs formulations while illumination revealed that there were some bacterial colonies on and under LSA\_AgNP/sCF formulation while there was no illumination for LAPA\_AgNP-sCF and L5AS\_AgNP-sCF formulations. In the case of Colistin functionalized AuNP-GC-sCFs, illumination did not provide any useful result due to the fact that excess *P. aeruginosa* growth (the fluorescent products spread around) (Fig. 9). In the case of *E. coli*, *S. epidermidis* and *K. pneumoniae*, LAPA\_AgNP-sCF and L5AS\_AgNP-sCF formulations suppressed all the growth while the AuNP-GC-sCFs did not provide 100% growth for Colistin resistant *K. pneumoniae* (Figure S9). Propagation of the rinsed NP-sCFs formulations from *P. aeruginosa* samples revealed that LSA\_AgNP-sCF, LAPA\_AgNP-sCF, L5AS\_AgNP-sCF, GODA\_AuNP-sCF-C, L5AS\_AuNP-sCF-C, and LAPA\_AuNP-sCF-C, LSA\_AuNP-sCF-C 12%, 5%, 1%, 1%, 1%, 2% and 4% bacterial growth in comparison to the control cotton fabric. So, this means that *P. aeruginosa* could penetrate into the formulated cotton fabrics and even form films. Interestingly, free colloidal LSA\_AgNPs showed higher antibacterial activity while the lowest antibacterial activity for the AgNP-sCFs were obtained for the

LSA\_AgNP-GC-sCF formulation, which could be related to that  $\text{Ag}^+$  release can be lessened by the formulation.

Fig. 10 reveals that GODA\_AuNPs interacted with *P. aeruginosa* while L5AS\_AuNPs did not give the interaction. GODA\_AuNPs and L5AS\_AuNPs possess +34 mV and -34 mV, respectively (the measurements were done triple using Zetasizer, Malvern Panalytical).

AgNPs are potentially toxic to microorganisms, which was shown by *in vitro* and *in vivo* studies (Baveloni et al., 2025). Colloidal AgNPs follow distinct mechanisms to prevent bacterial growth including release of  $\text{Ag}^+$  ion, penetration into cytoplasm that results in oxidative stress, disintegration of cellular membrane's integrity and interfering with key proteins and nuclear material (Bruna et al., 2021). The AgNPs impregnated onto the cotton fabric relies on  $\text{Ag}^+$  release (Tahir et al., 2024) that can cause DNA damage and reactive oxygen species (ROS) production within bacterium (e.g. *P. aeruginosa*) (Linh et al., 2022). Besides, the hydrophobicity of AgNP-cotton fabric is also critical for bacterial attachment to the surface that also brings an indirect antibacterial activity (Linh et al., 2022). It should be noted that bacterial species can gain resistance to AgNPs (Jain et al., 2022) while the differences between the antibacterial efficacy of AgNP-Cotton fabrics for different bacterial species are not related to the any resistance, rather it is related to the sugar ligand used in the synthesis of AgNPs. Therefore, it is possible to speculate that capture of the bacterial cells on the cotton fabrics through carbohydrate-lectin interactions played roles in addition to the release of  $\text{Ag}^+$  ions. In contrast to the AgNP-cotton fabrics, Colistin functionalized AuNP-sCFs brought out the major antibacterial activity by the presence of Colistin. Drastic differences between the bacterial responses to the Colistin functionalized sCFs could be related to two major parameters including (i) the degree of Colistin resistance of the bacterial species and (ii) bacterial interaction with the carbohydrate



**Fig. 9.** Antibacterial activity of AgNPs (a) and nanoparticle enhanced cotton fabric (b). i: Control; ii: LSA\_AgNP-sCF; iii: LAPA\_AgNP-sCF; iv: L5AS\_AgNP-sCF; v: GODA\_AuNP-sCF; vi: L5AS\_AuNP-sCF; vii: LAPA\_AuNP-sCF; viii: LSA\_AuNP-sCF and + refers to the presence of Colistin (100 ng/cm<sup>2</sup>) (C).

**Table 1**

MIC and MBC values of the AgNPs on *P. aeruginosa*, *E. coli*, *S. epidermidis* and *K. pneumoniae*.

Nanoparticle	MBC/MIC values ( $\mu\text{g/mL}$ )			
	<i>P. aeruginosa</i>	<i>E. coli</i>	<i>S. epidermidis</i>	<i>K. pneumoniae</i>
L5AS_AgNP1	0.634/0.634	0.634/ 0.634	0.634/0.634	0.317/0.317
L5AS_AgNP4	Not available	Not available	Not available	Not available
LSA_AgNP5	0.078/0.039	0.078/ 0.039	0.078/0.039	0.078/0.078
LAPA_AgNP1	0.634/0.634	0.634/ 0.634	Not available/ 0.634	0.634/0.317
LSA_AgNP1	0.158/0.079	0.158/ 0.158	0.317/0.158	0.317/0.158

derivative on the surface. Overall, this study shows that decoration of fabric surfaces with carbohydrate derivative synthesized AgNPs or antibiotic functionalized AuNPs could bring semi-selective and strong antibacterial activity.

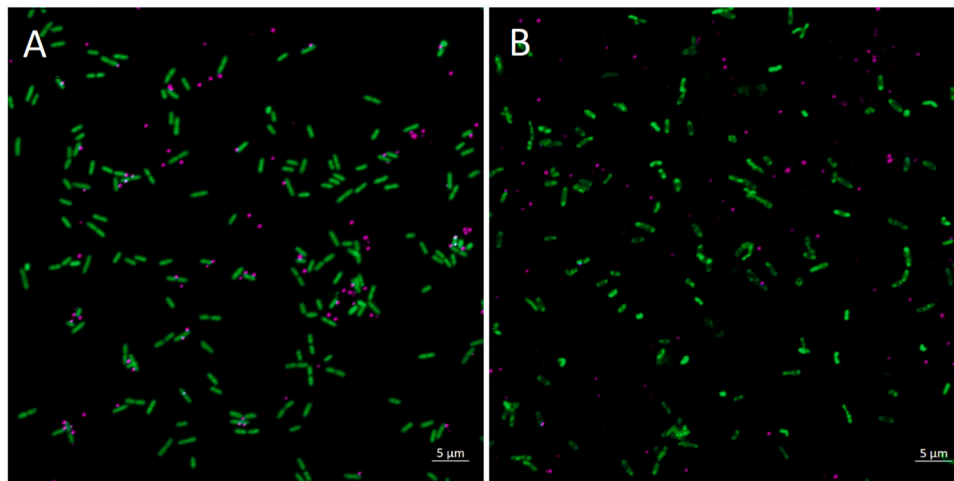
It is also noteworthy to mention that this study has some limitations including (i) Ag<sup>+</sup> ion release from AgNPs synthesized with different sugar ligands and (ii) formation of silver oxide phases on the metallic AgNPs that might affect the antibacterial activity were not characterized. Besides (iii) wettability of the surfaces (which can affect bacterial attachment), (iv) SEM imaging of bacteria treated cotton fabric formulations to show capture of bacterial cells and formation of bacterial biofilms and (v) the degree of Colistin resistance for each bacterial cells need characterization to provide a deeper understanding of the observed semi-selective antibacterial activity.

#### 4. Conclusion

*In situ* synthesis of metallic AgNPs and AuNPs on/in scoured cotton fabrics provide homogeneous surfaces in comparison to that administration of the nanoparticles on scoured cotton fabric either predated with Cysteine or not. Silver nanoparticle glycoconjugates and *in situ* synthesized AgNP-GC-sCFs and Colistin functionalized AuNP-GC-sCFs showed excellent anti *P. aeruginosa* activity, and similarly colloidal AgNP-GCs showed none-selective strong antibacterial activity for gram (-) *E. coli*, *K. pneumoniae* and gram (+) *S. epidermidis*. The study not only shows AgNP-GCs on sCFs can be highly noxious to pathogens, but also it shows Colistin functionalization on AuNP-GCs on sCFs can be toxic to the antibiotic-resistant species. So, these findings can call for further research on surface design using carbohydrate ligands that can target bacterial species through their surface lectins in order to overcome resistance in the development of biomedical coating materials. Besides it is also critical to understand how exactly the surfaces played roles for the semi-selective antibacterial activity, which should be paid attention since biomedical coating materials are critical in order to prevent hospital acquired infections.

#### CRediT authorship contribution statement

**Kırbay Fatma Öztürk:** Investigation, Funding acquisition, Formal analysis, Data curation. **Küçük Ardanur:** Writing – original draft, Investigation, Funding acquisition, Formal analysis. **Hafize Dilek Tepe:** Investigation, Formal analysis. **Kasemets Kaja:** Writing – review & editing, Supervision, Investigation, Funding acquisition, Conceptualization. **Sancak Sedanur:** Writing – original draft, Investigation, Formal analysis. **Otsus Maarja:** Methodology, Investigation, Formal analysis. **Güney Şevval:** Investigation, Formal analysis, Data curation. **Sel**



**Fig. 10.** Confocal laser scanning microscopy image of GODA\_AuNPs (A) and L5AS\_AuNPs (B) interactions with bacteria *P. aeruginosa*. Cells were stained by green fluorescent stain Syto 9, and AuNPs were visualized by reflection mode at 640 nm (purple dots). All scale bars are 5  $\mu$ m.

**Sedanur:** Investigation, Formal analysis, Data curation. **Kılınc Çetin:** Writing – review & editing, Validation, Supervision, Software, Formal analysis, Conceptualization. **Taşdelen Zehra:** Investigation, Formal analysis, Data curation. **Esinti İrem Demirbaş:** Investigation, Formal analysis, Data curation. **Esinti İdris:** Writing – review & editing, Writing – original draft, Supervision, Investigation, Funding acquisition, Formal analysis, Data curation, Conceptualization.

#### Declaration of Competing Interest

The authors declare no conflict of interest.

#### Acknowledgements

The authors acknowledge TÜBİTAK 2209 Student Projects of Türkiye (Grant number 1919B012109406), Kastamonu University Internal Grant (Türkiye, Grant number KÜBAP-01/2022–20), the COST Action CA21145-(EURESTOP) (European Cooperation in Science and Technology, European Network for diagnosis and treatment of antibiotic-resistant bacterial infections, Bruxelles-Belgium), Fatma Öztürk Kırbay personally thanks to 2218 TUBITAK National Postdoctoral Research Fellowship Program 2021/2 of Türkiye, and Estonian Research Council Program PRG749.

#### Appendix A. Supporting information

Supplementary data associated with this article can be found in the online version at [doi:10.1016/j.microb.2025.100269](https://doi.org/10.1016/j.microb.2025.100269).

#### Data availability

Data will be made available on request.

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